

Net vs Gross Primary production

Energy source:
1,254,000
kcal/m²/year

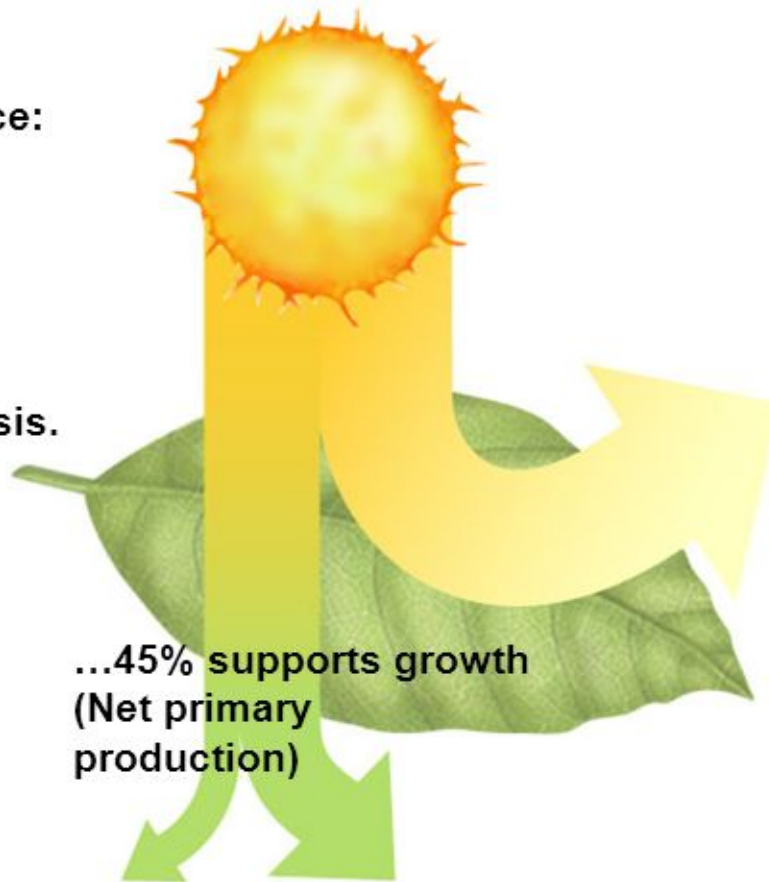
0.8% energy
captured by
photosynthesis.
Of this...

...55% lost
to respiration

...45% supports growth
(Net primary
production)

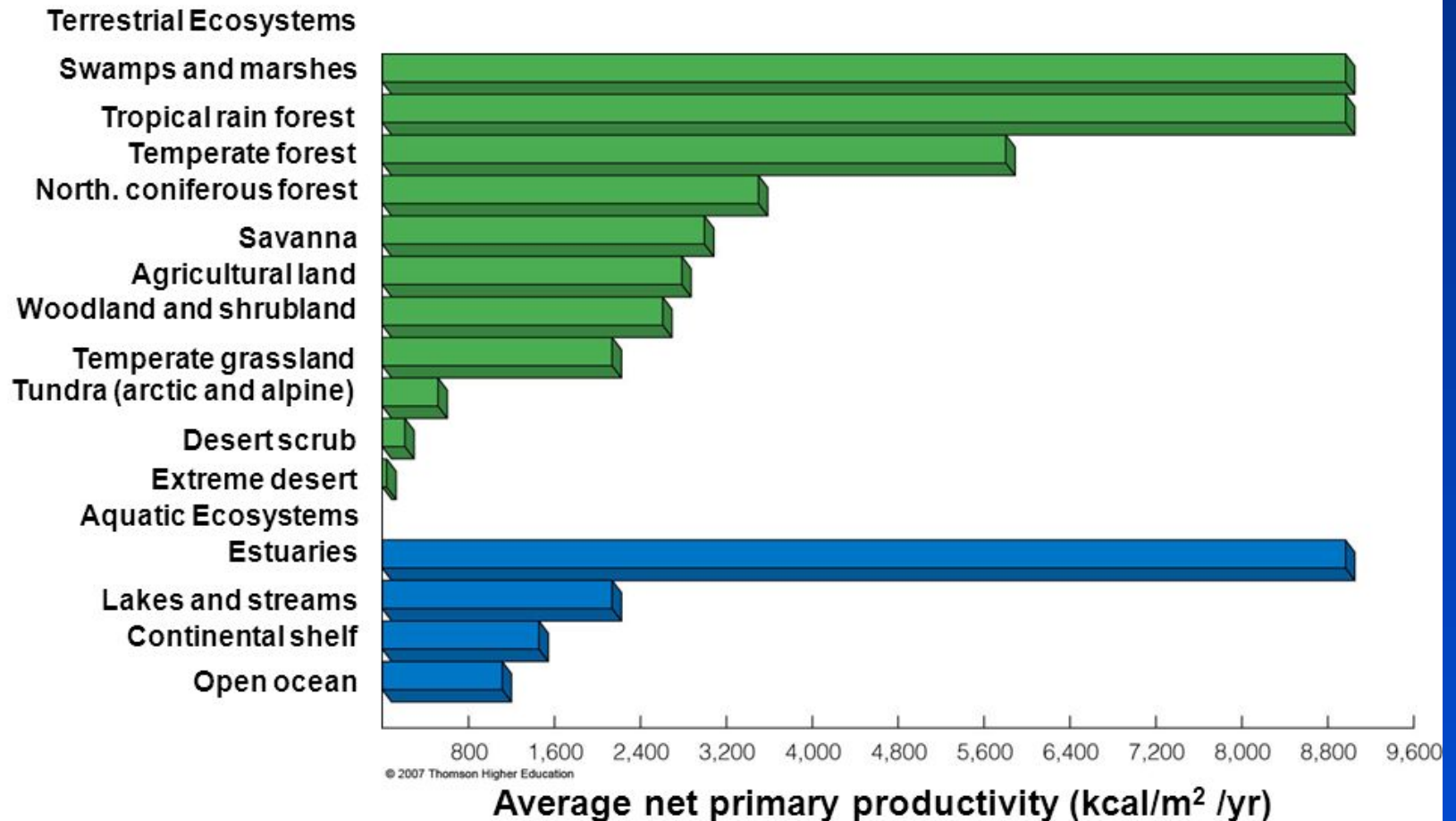
...11% enters
grazing food web

...34% enters
decomposer food web
as dead material

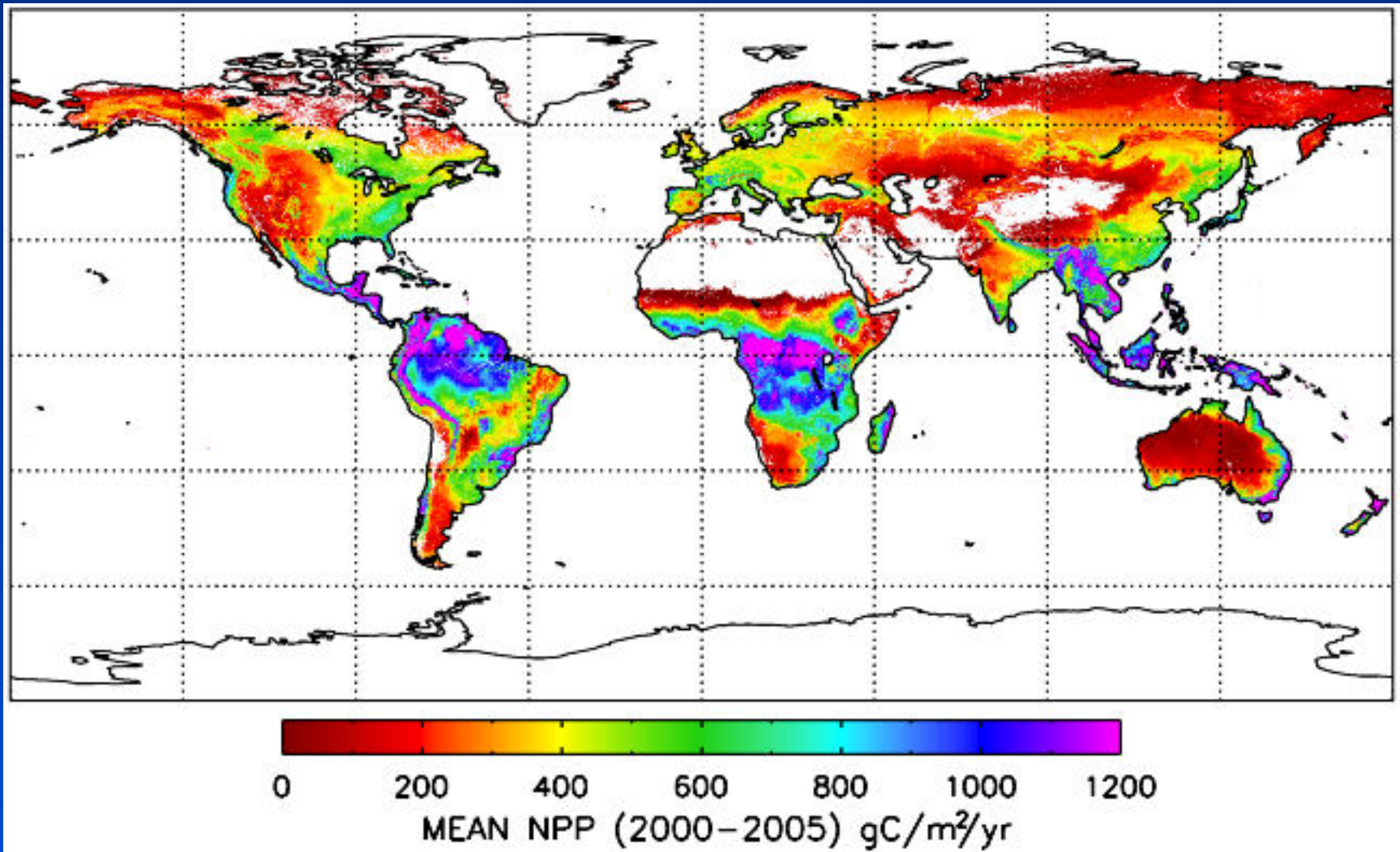


Net primary productivity

Terrestrial vs marine ecosystems



Net primary productivity (annual average) Terrestrial systems



Primary production in terrestrial systems

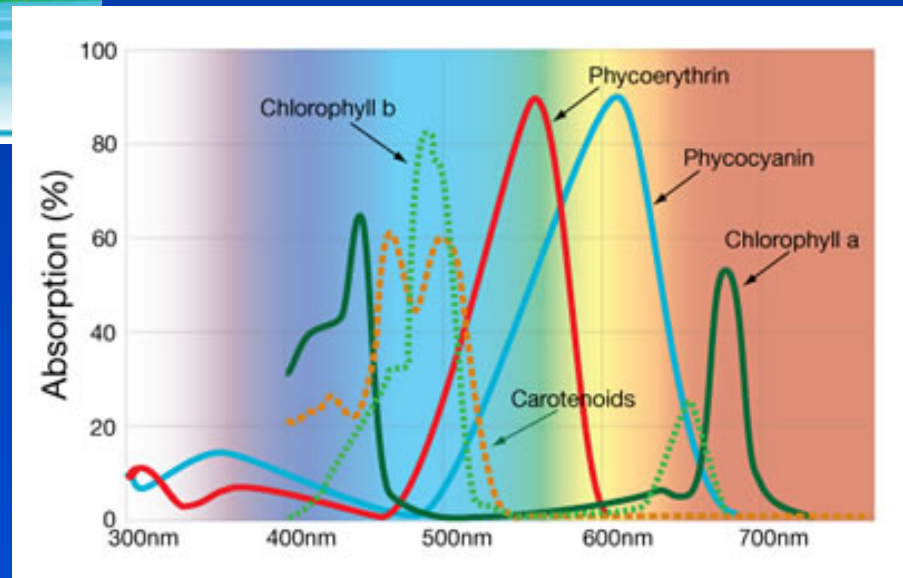
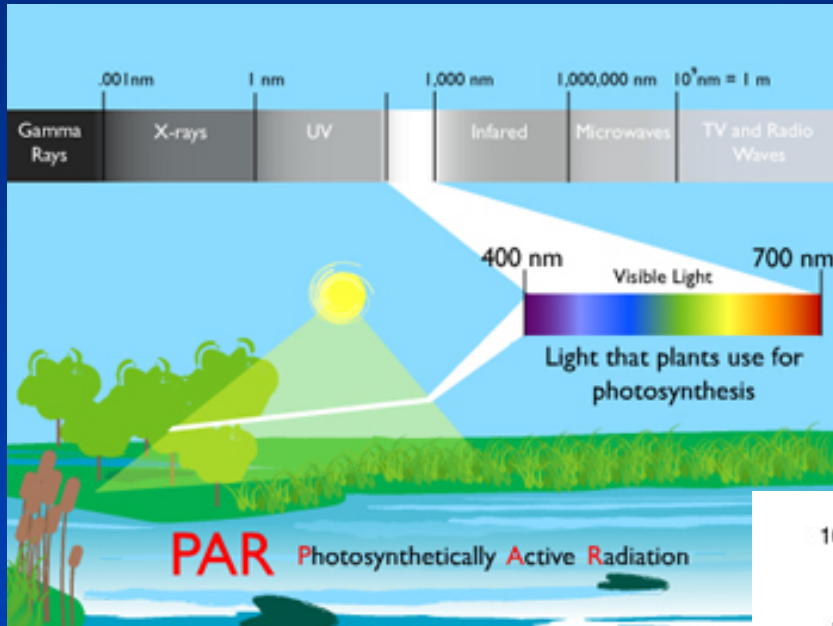
PP in terrestrial ecosystems is performed by autotrophic plants through photosynthetic processes

C3, C4, CAM etc.

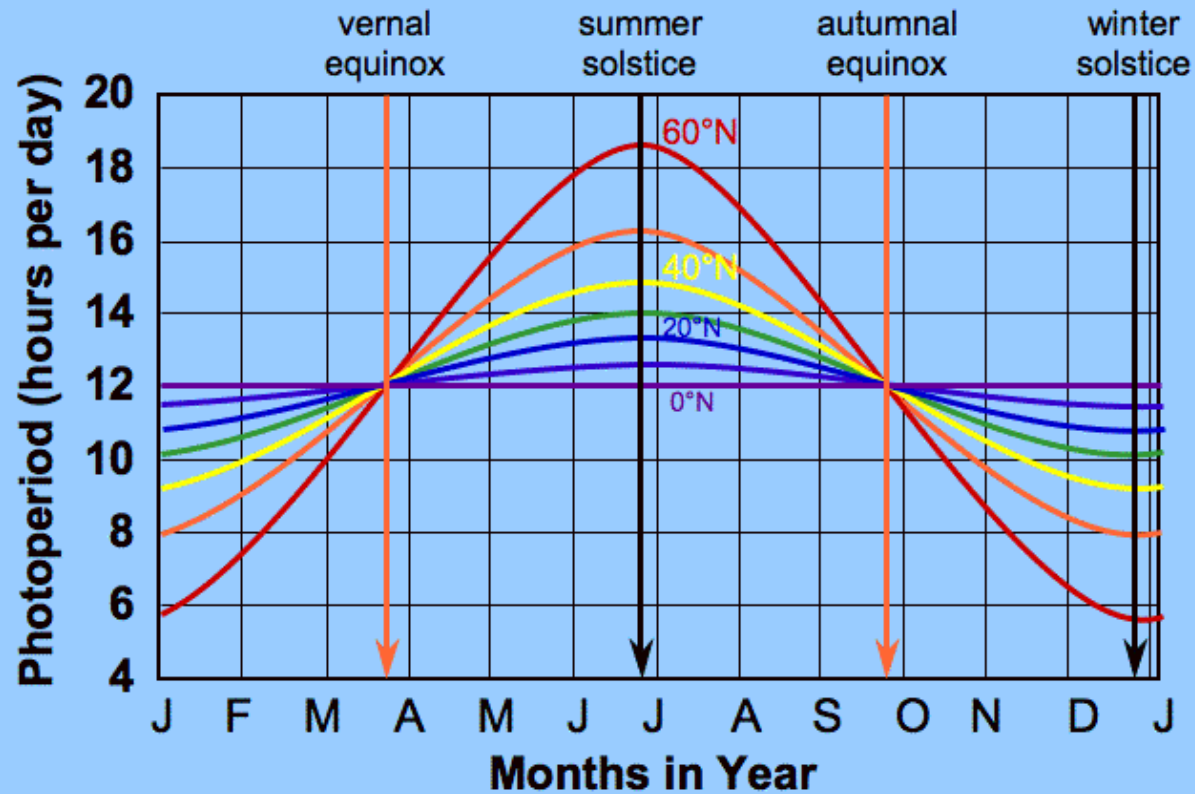
Principal factors affecting terrestrial PP are:

- ✓ Light availability (season/latitude)
- ✓ Nutrient availability
(soil composition, recycling)
- ✓ Water availability (precipitation, soil physical characteristics)
- ✓ Disturb such as fire, grazing (natural and anthropogenic)

Photosynthetically Active Radiation



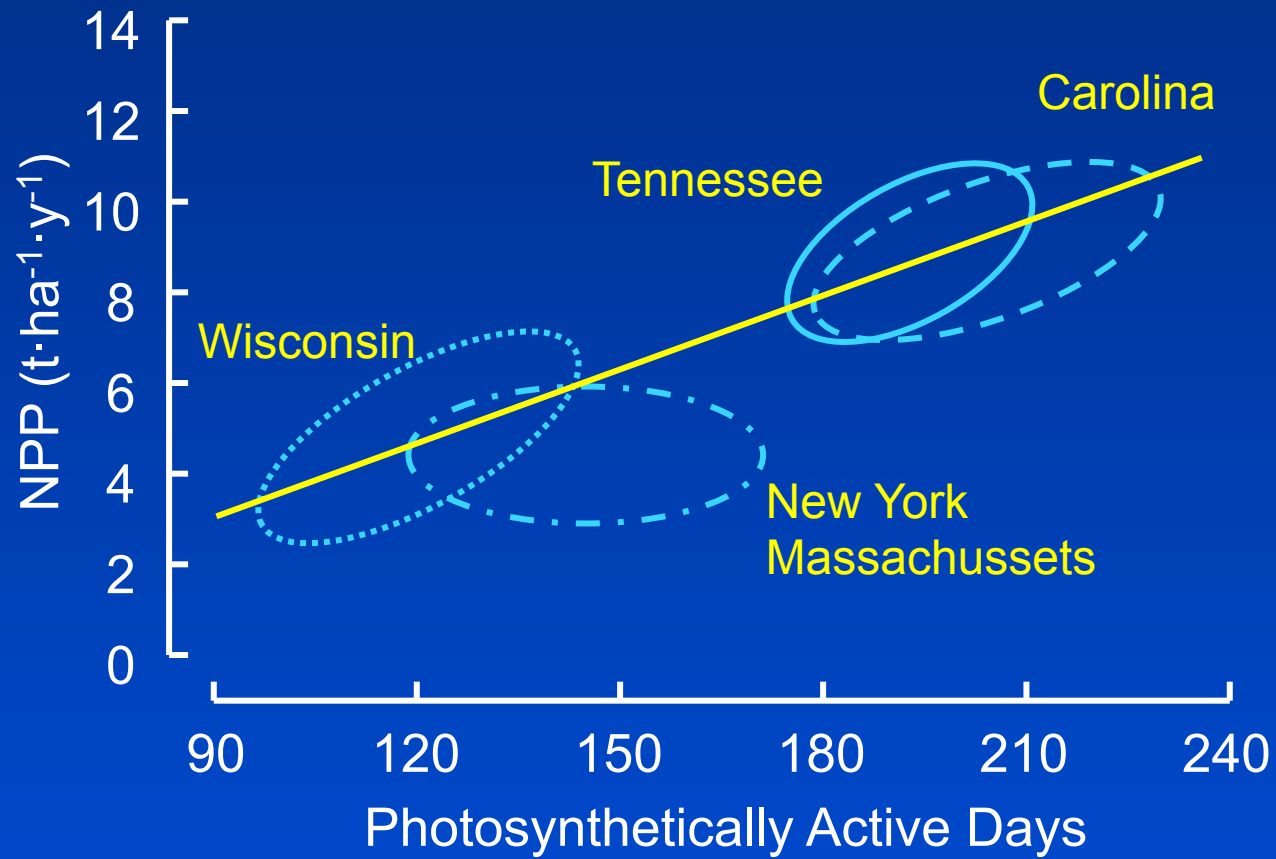
Photoperiod length



Willimantic, CT: 41.7° N 72.2° W

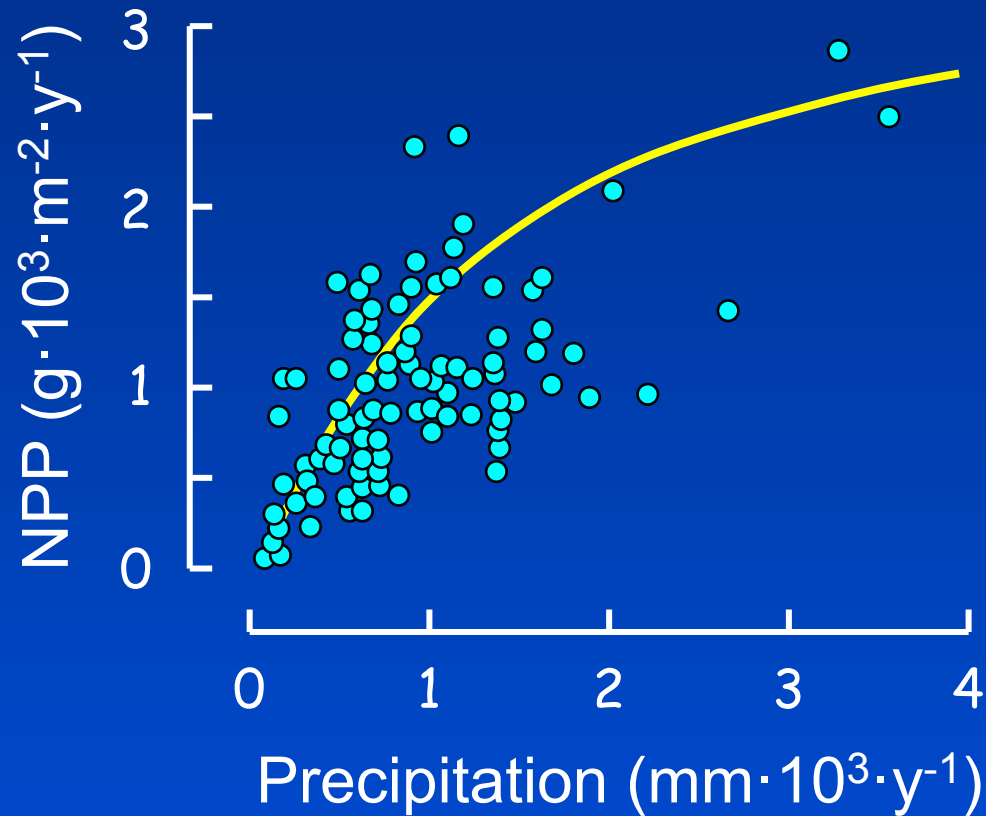
Net Primary Productivity in temperate forests

Central-East USA



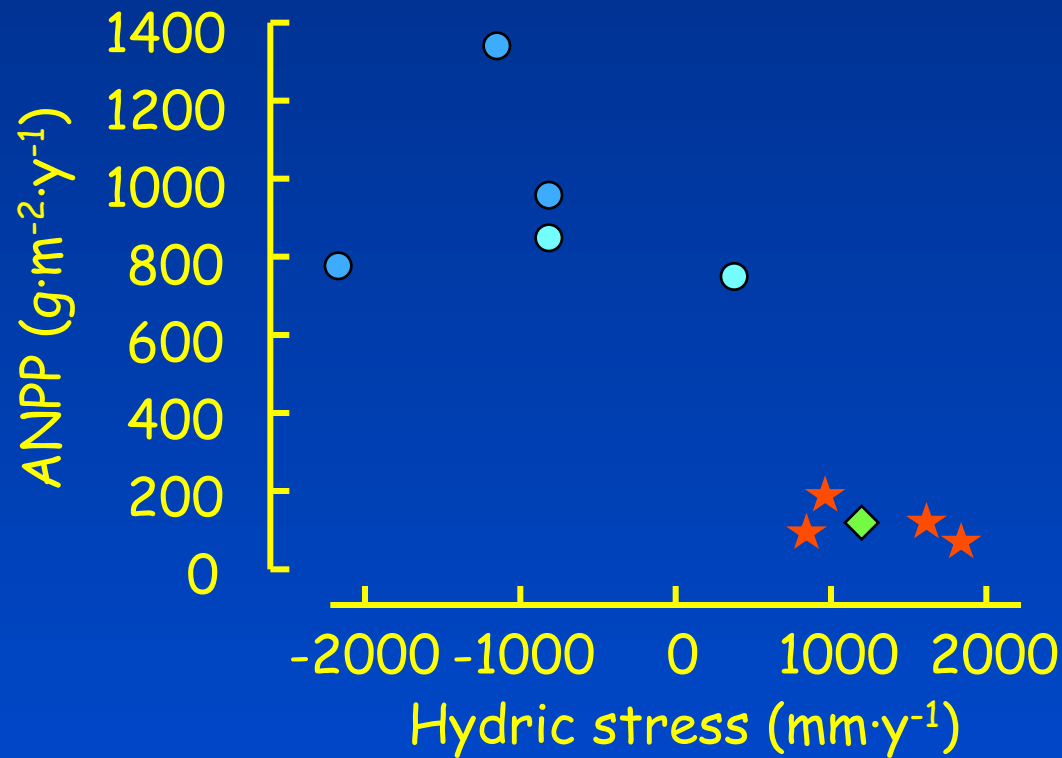
NPP relationship with precipitations

Temperate forests Europe and USA



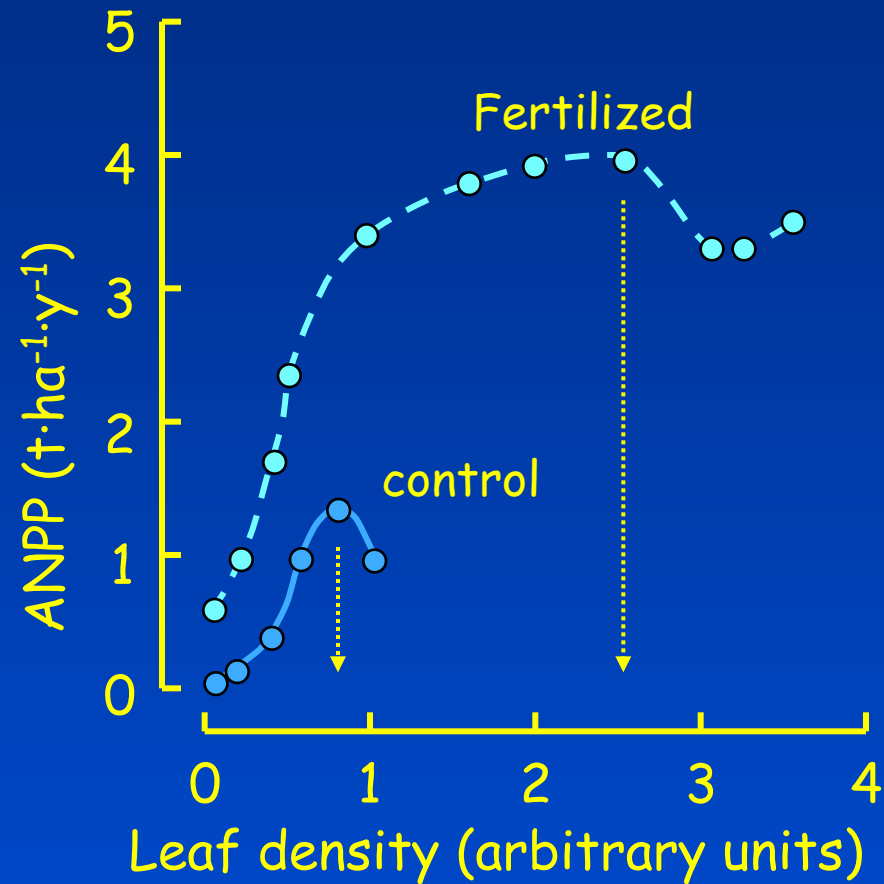
NPP relationship with hydric stress

Temperate forests Europe and USA



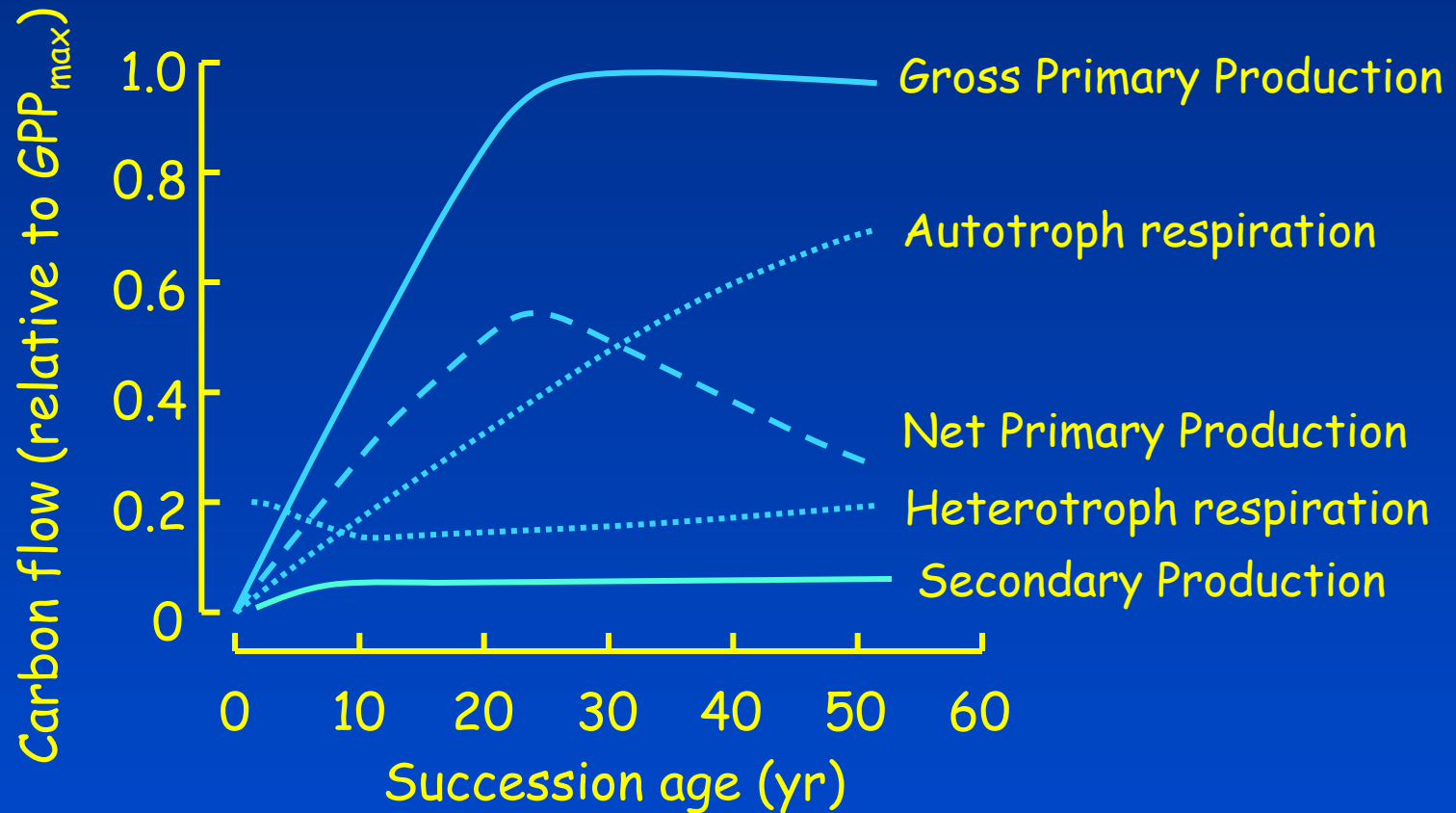
NPP increase with soil fertilization

Chestnut stand

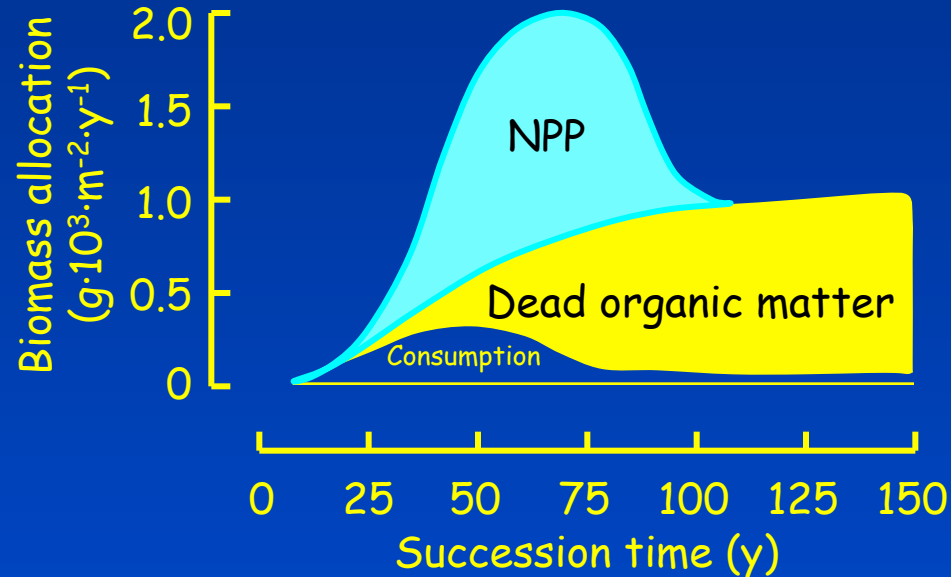


Production and respiration along the succession

in an evergreen temperate forest (schematic)



Biomass allocation along the succession in a broadleaf temperate forest (schematic)

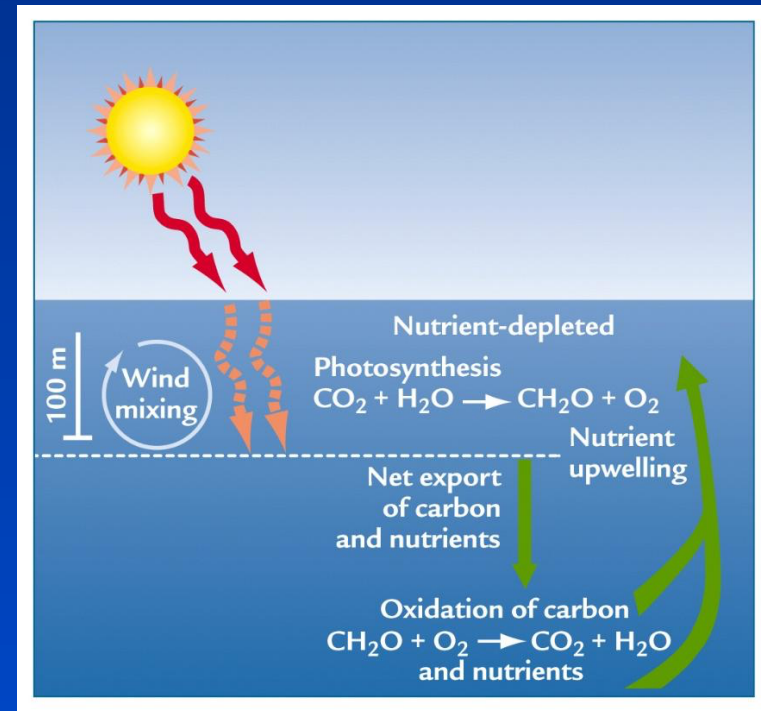


Primary production in oceanic systems

Most of oceanic PP is from photosynthetic phytoplankton operating at the ocean surface in the euphotic zone (ten to hundreds m depth)

Factors affecting oceanic PP are:

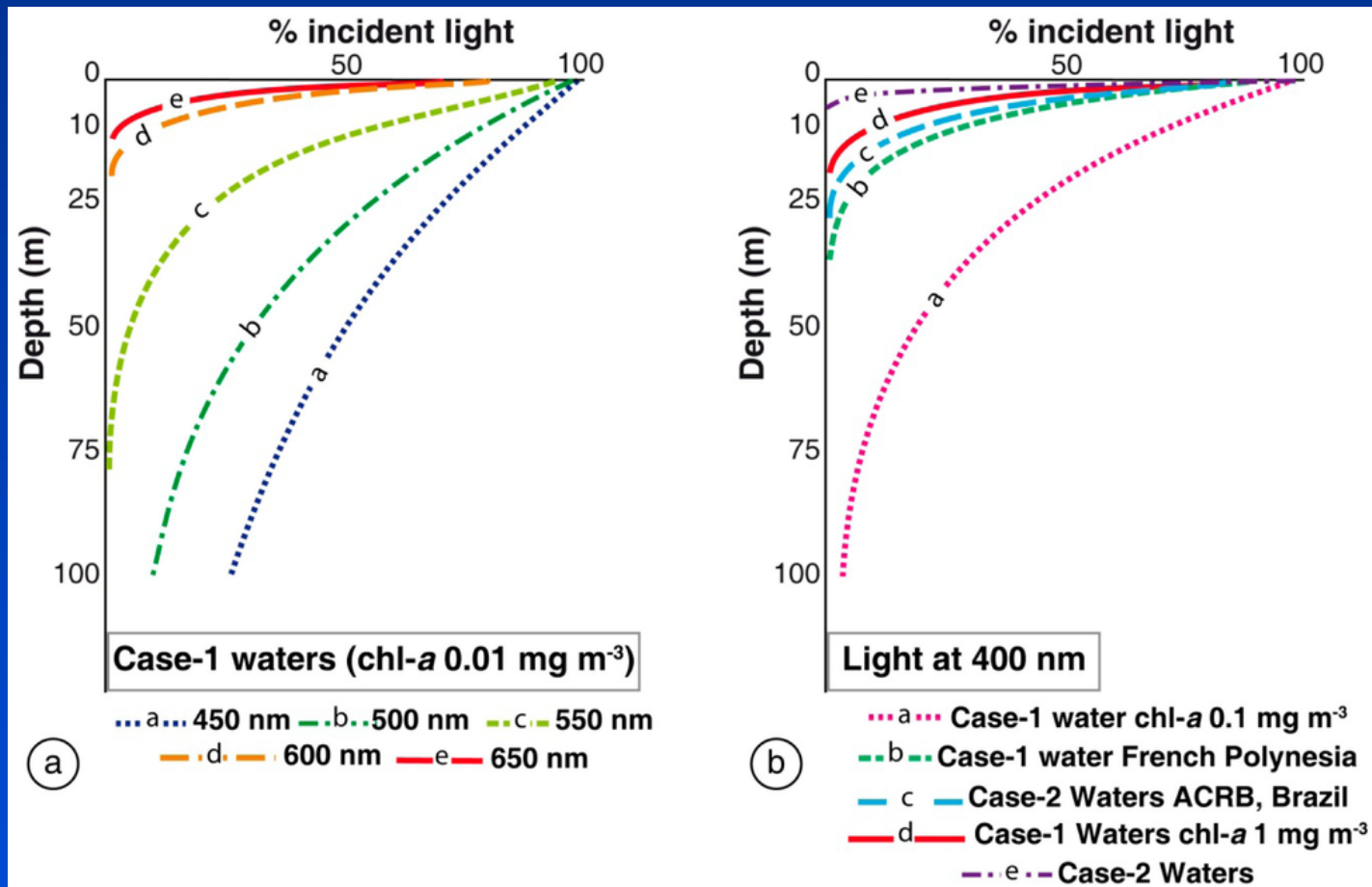
- ✓ Light availability (depth)
- ✓ Nutrient availability (recycling, runoff and upwelling)
- ✓ Temperature
- ✓ Water acidity (CO_2 dissolved by antropogenic release in the atmosphere)



Light extinction with dept in water

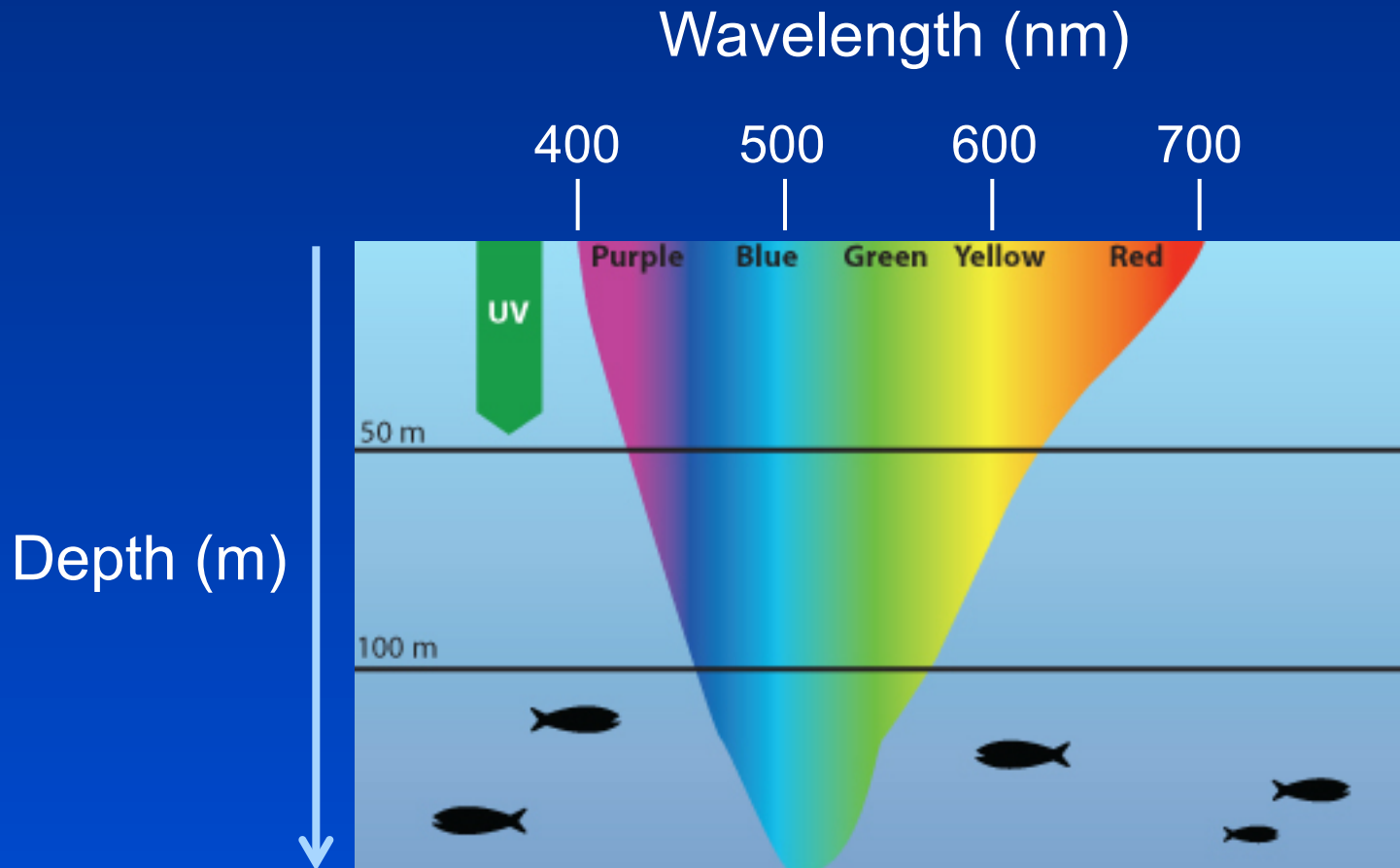
Lambert-Beer equation

$$I_z = I_0 \cdot e^{-kz} \quad (Z, \text{extinction coefficient})$$



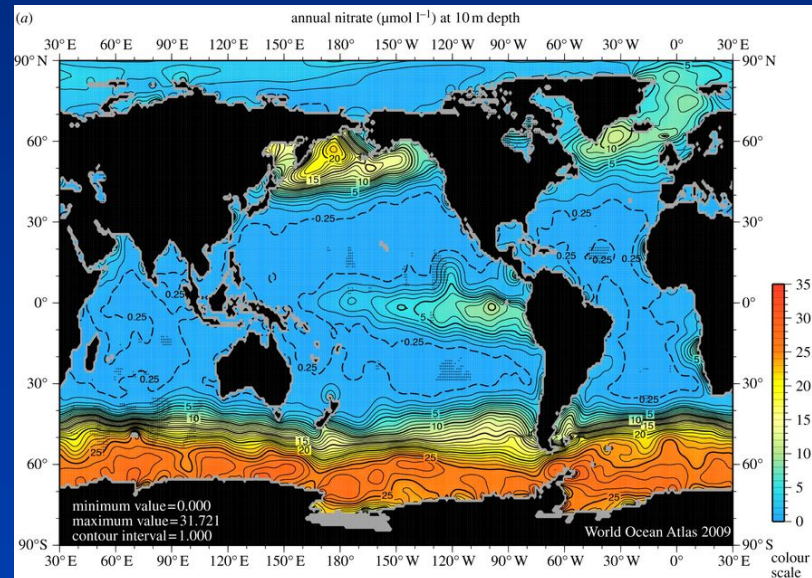
Light extinction with depth in water

Spectral variation

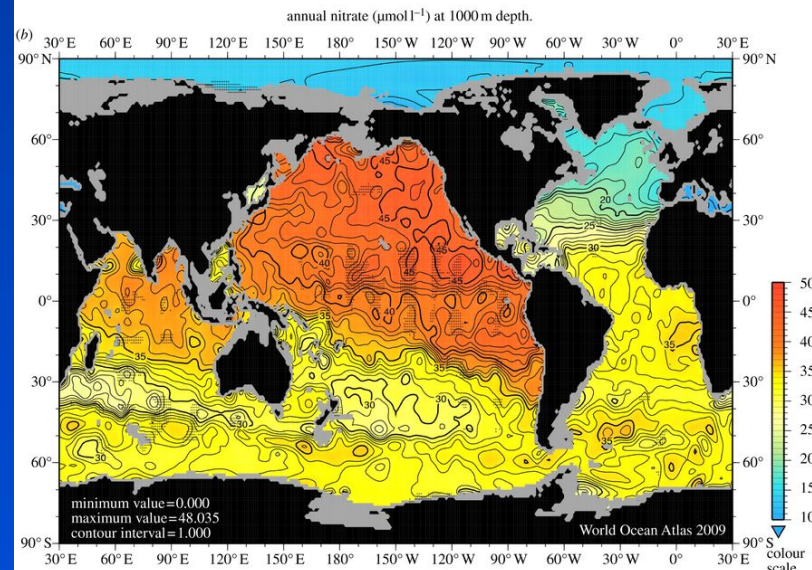


Nitrates dissolved in oceanic waters

Surface

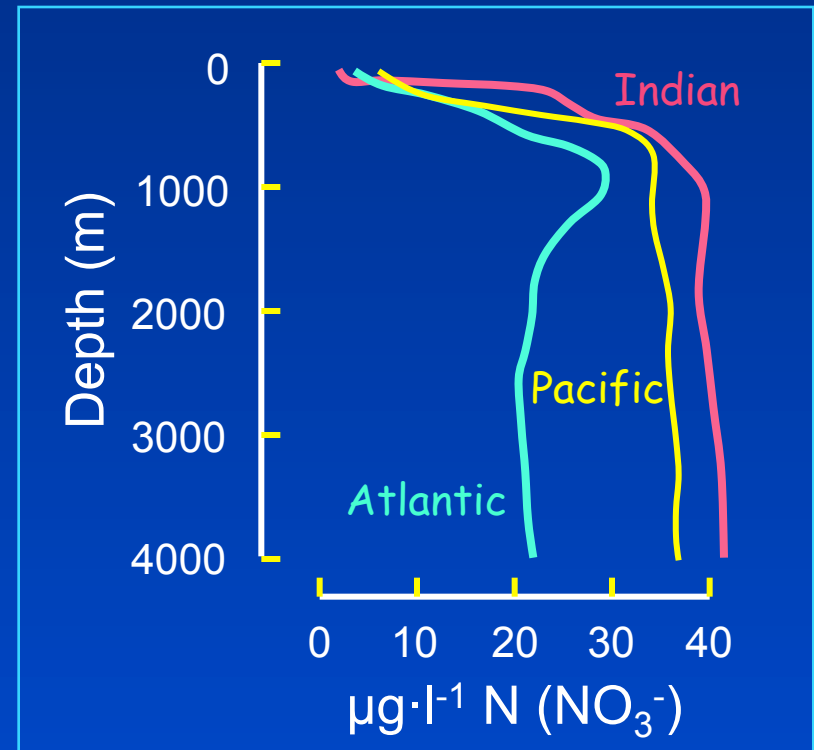
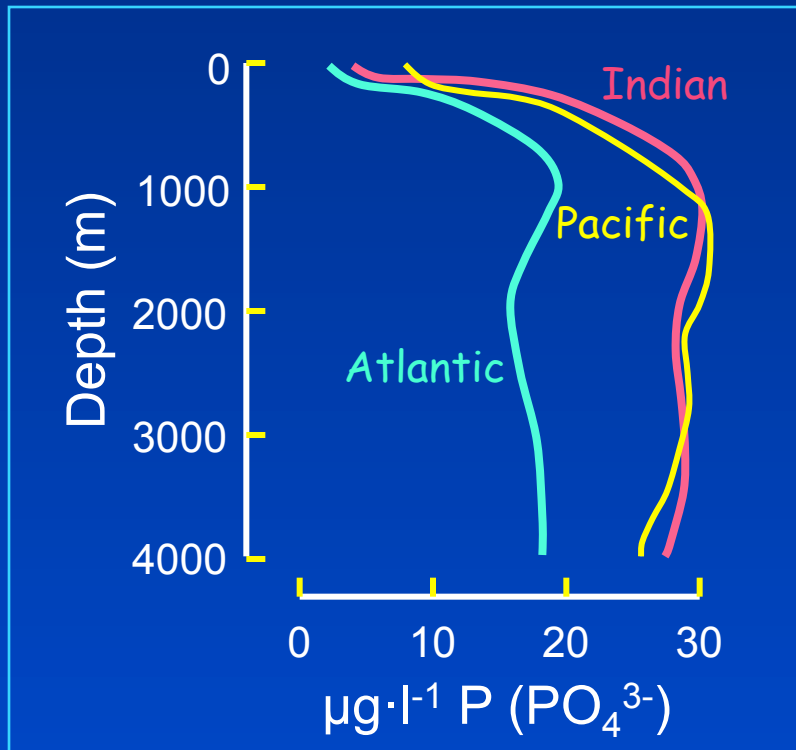


1000 m depth

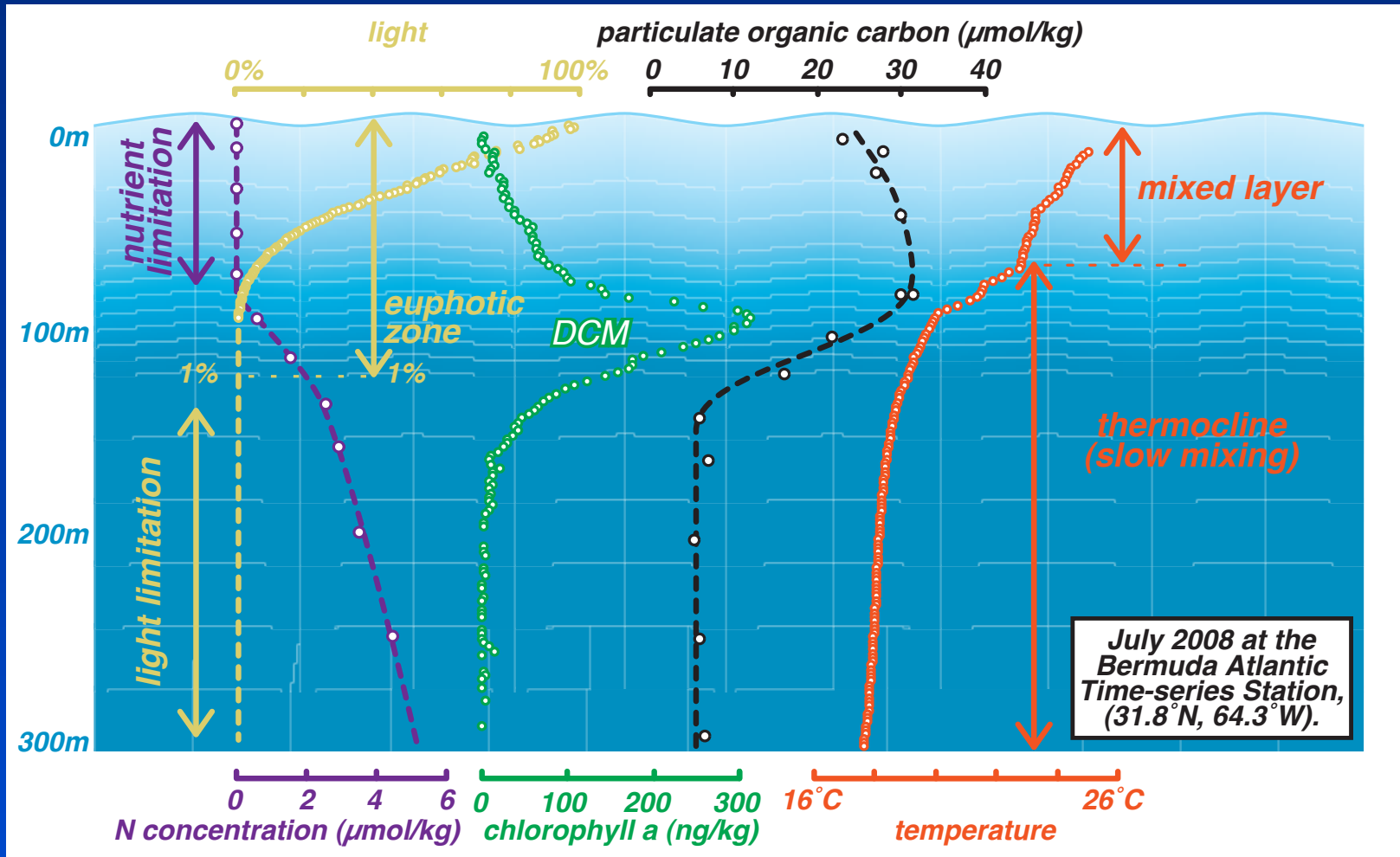


Phosphates & nitrates concentration

Oceanic waters

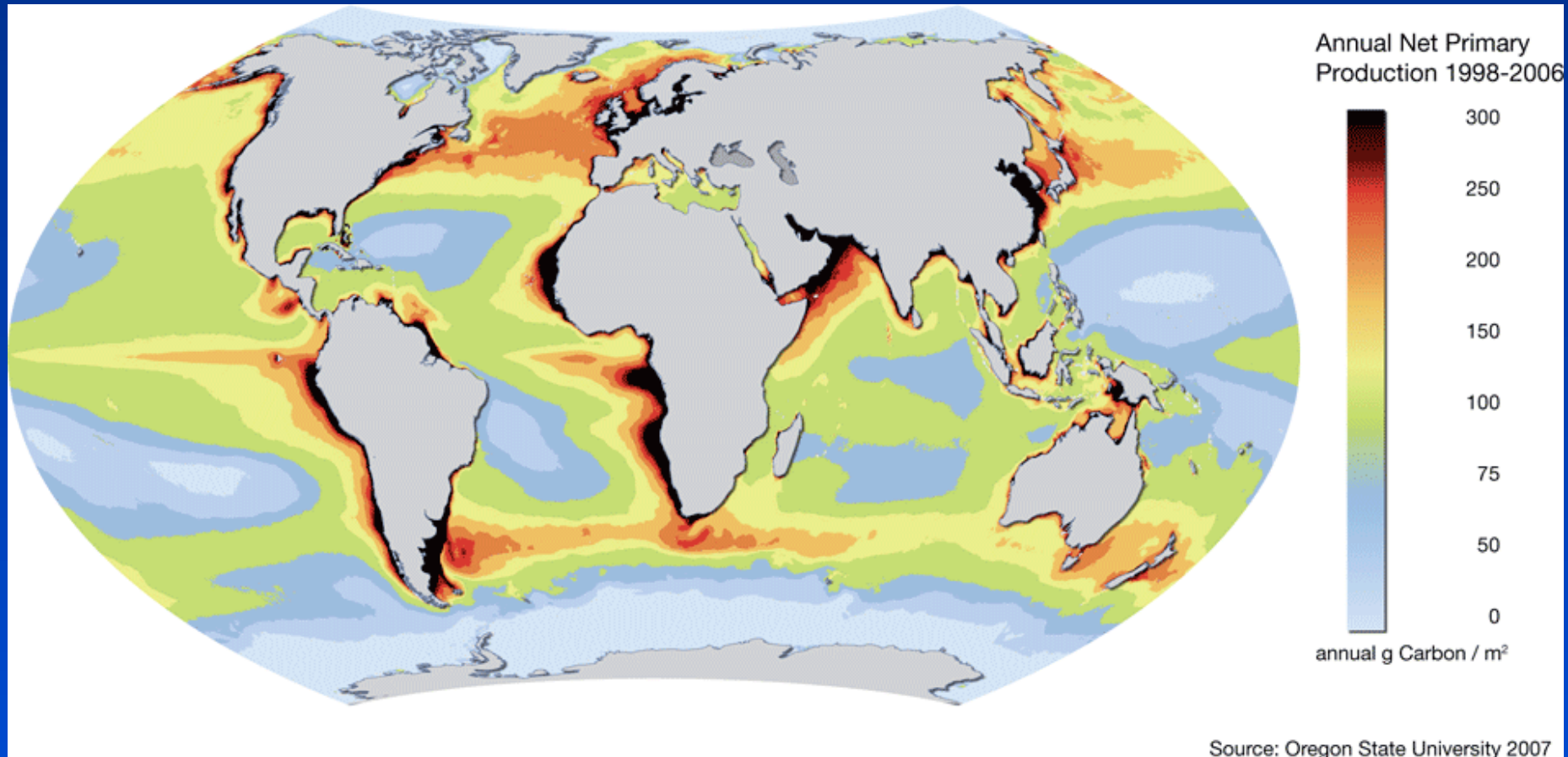


Physical/chemical gradients at ocean surface



DCM deep chlorophyll maximum

Net primary productivity (annual average) Oceanic systems

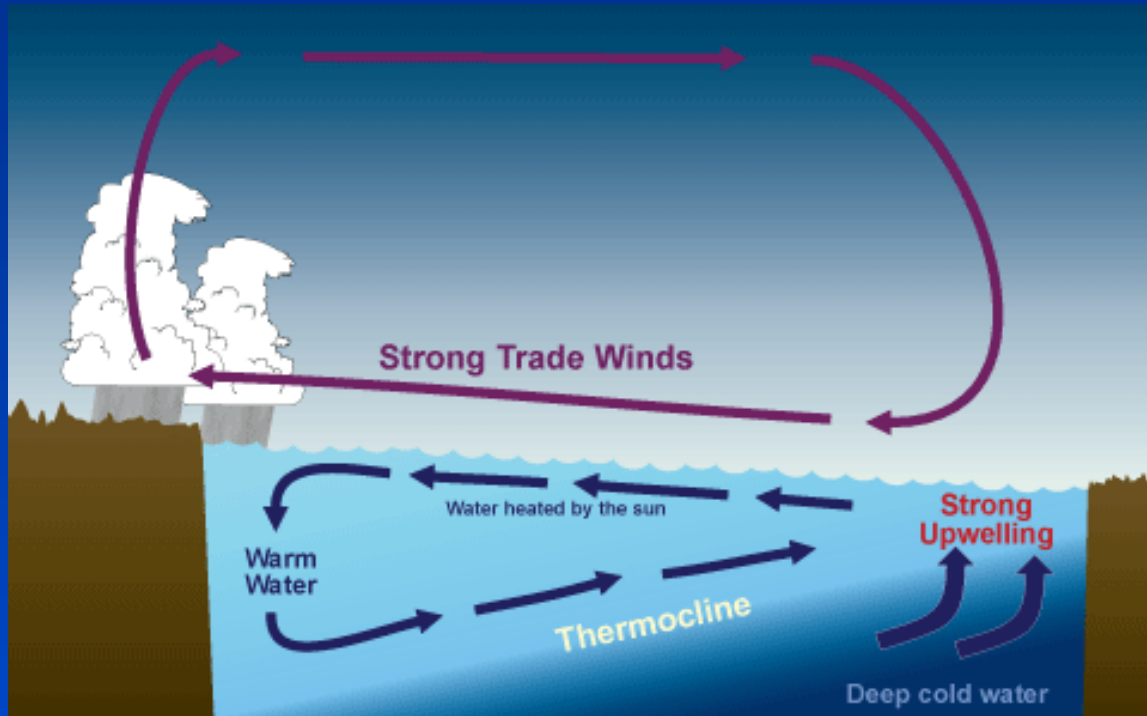


Source: Oregon State University 2007

Upwelling in equatorial Pacific

Low pressure

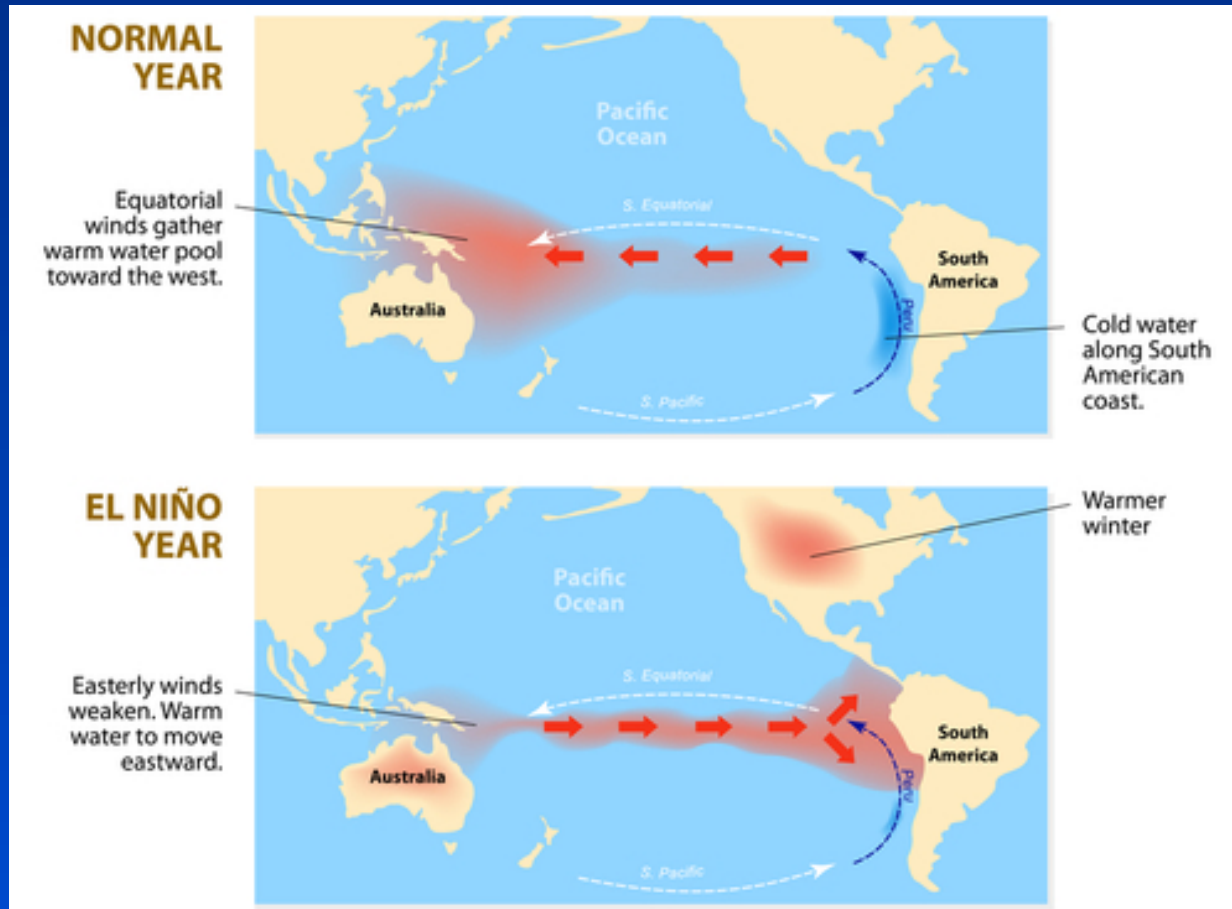
High pressure



Australia

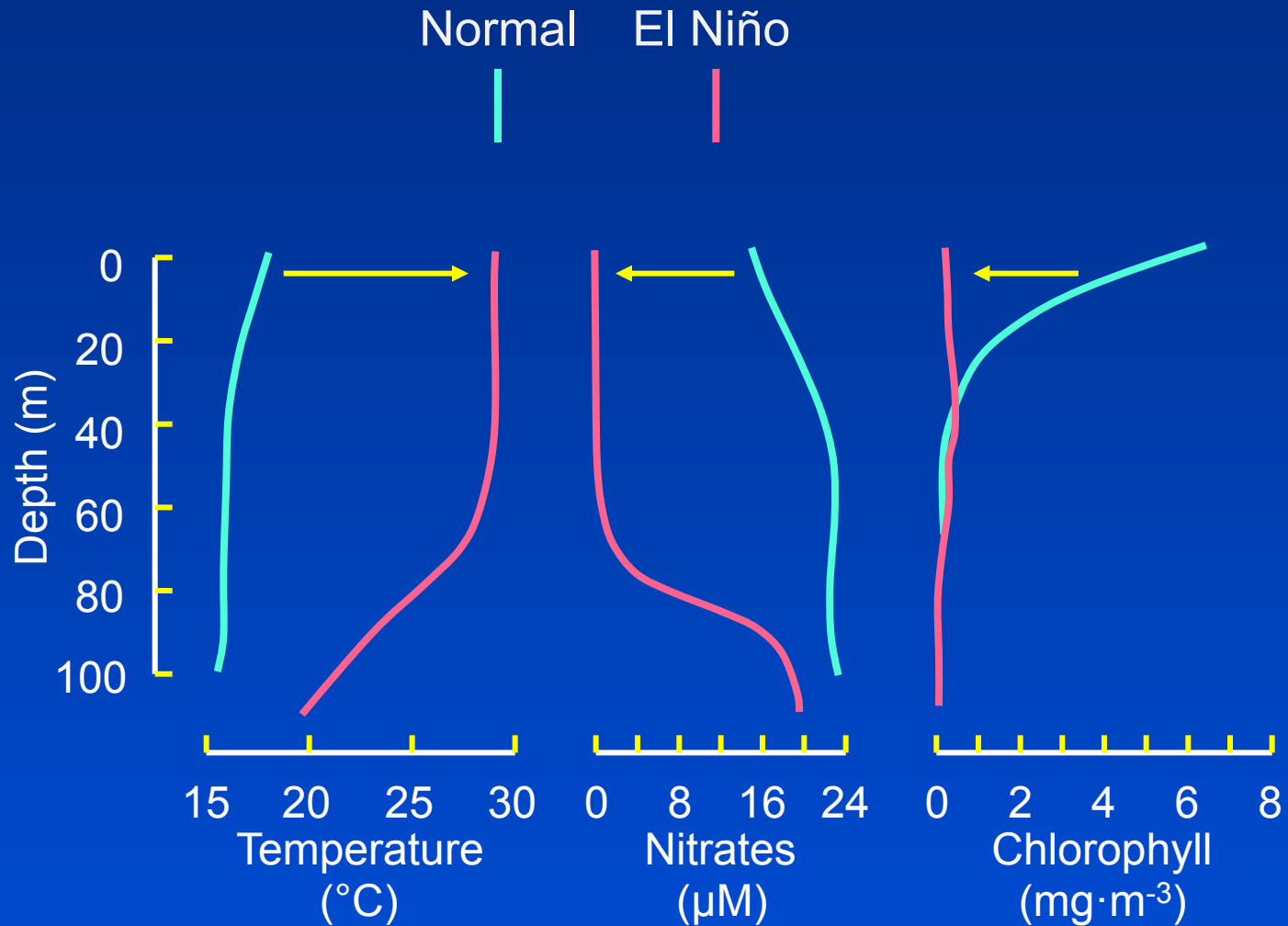
Peru

El Niño Southern Oscillation

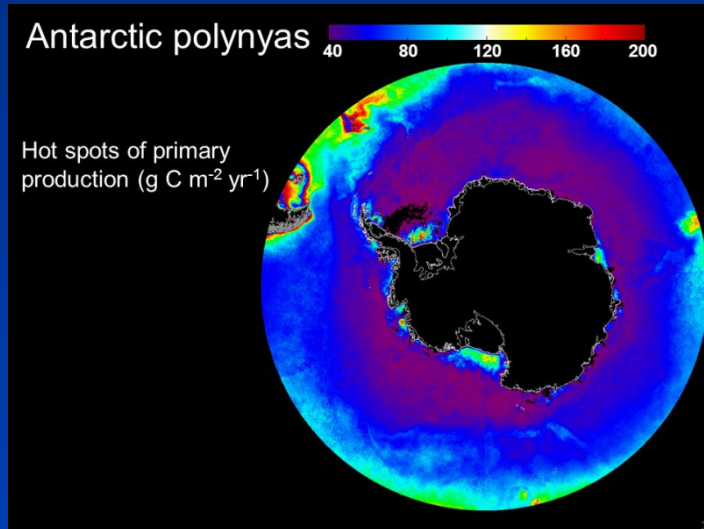


El Niño Southern Oscillation

Variations off the Peru coast

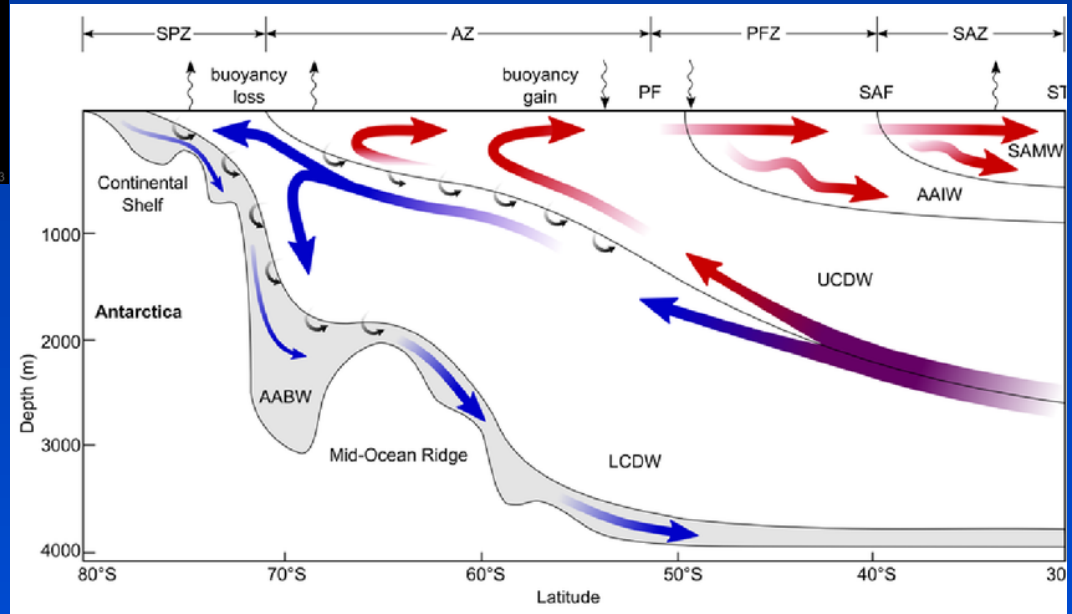


Circum-Antarctic overturning circulation



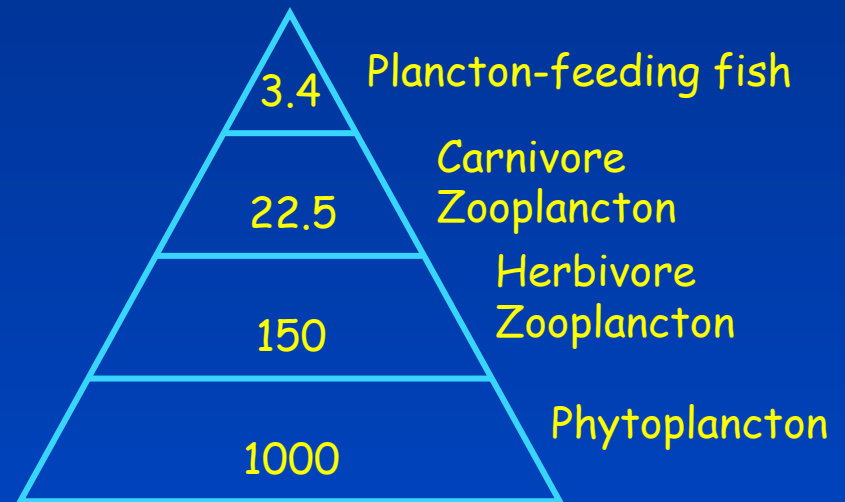
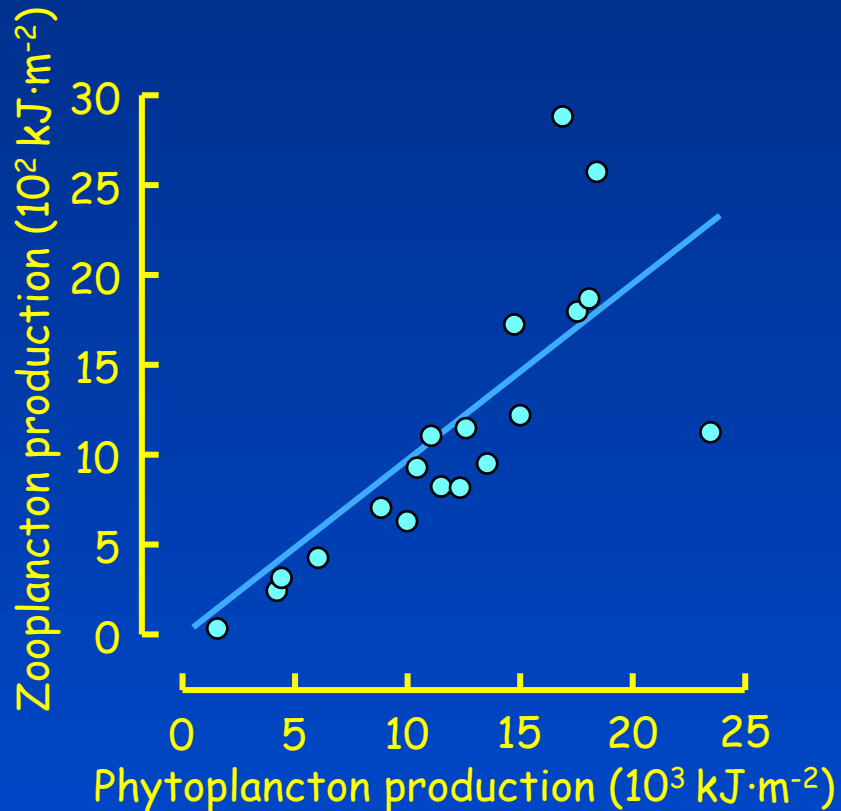
Primary productivity

STF, Sub-Tropical Front; SAF, Sub-Antarctic Front; PF, Polar Front; SAMW, Sub-Antarctic Mode Water; AAIW, Antarctic Intermediate Water; UCDW, Upper Circumpolar Deep Water; LCDW, Lower Circumpolar Deep Water; AABW, Antarctic Bottom Water; SAZ, Sub-Antarctic Zone; PFZ, Polar Frontal Zone; AZ, Antarctic Zone; SPZ, Sub-Polar Zone.

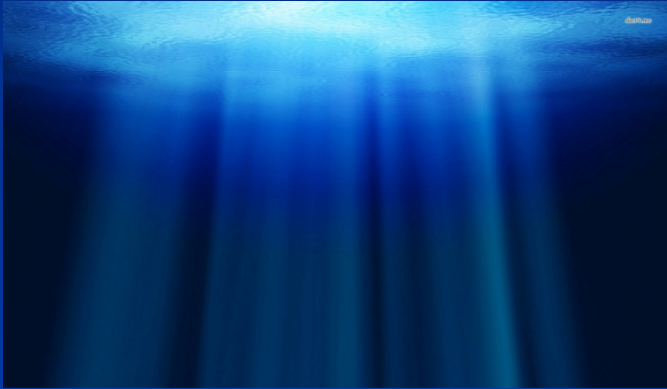


Oceanic trophic chain

Relative energy content of different trophic levels



Source-sink coupled communities



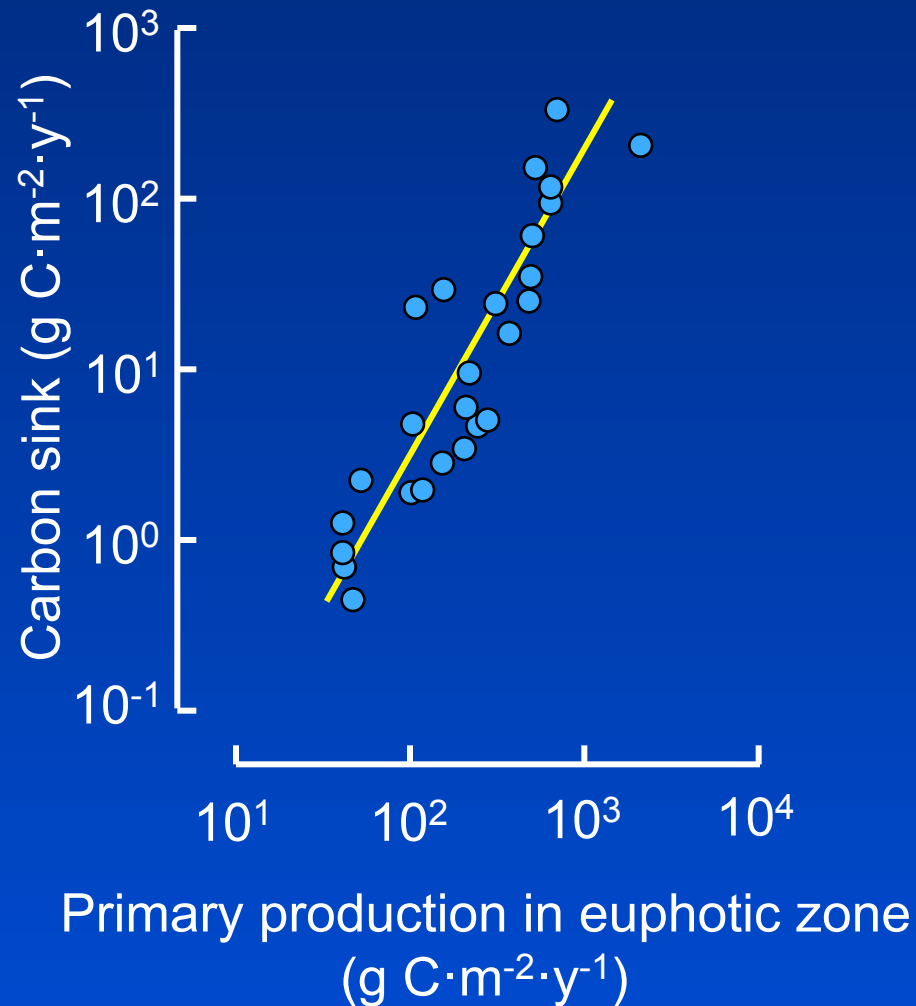
Euphotic zone

Gravity  Organic carbon "snow"

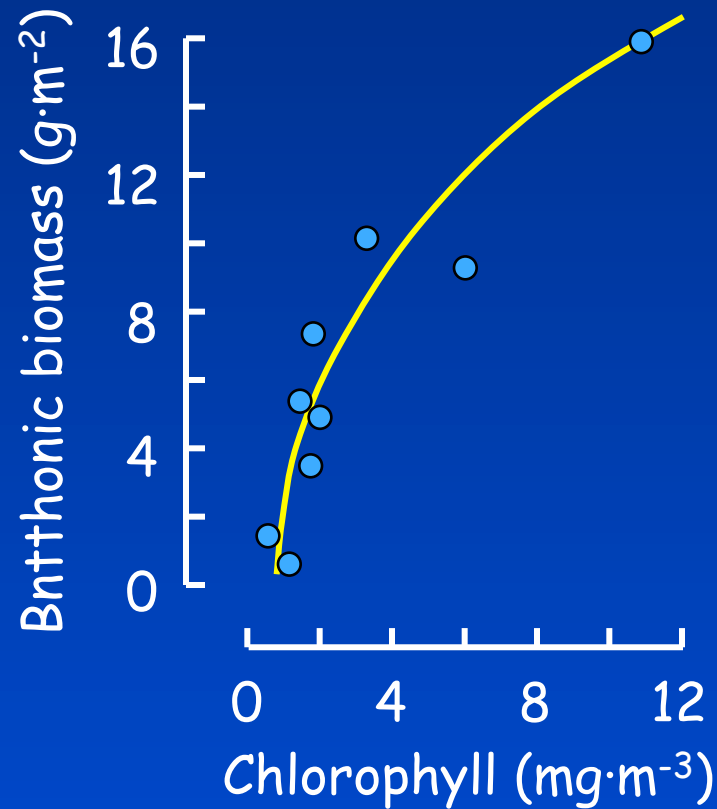


Benthic zone

Carbon sink from euphotic zone



Relationship between euphotic PP and benthonic production

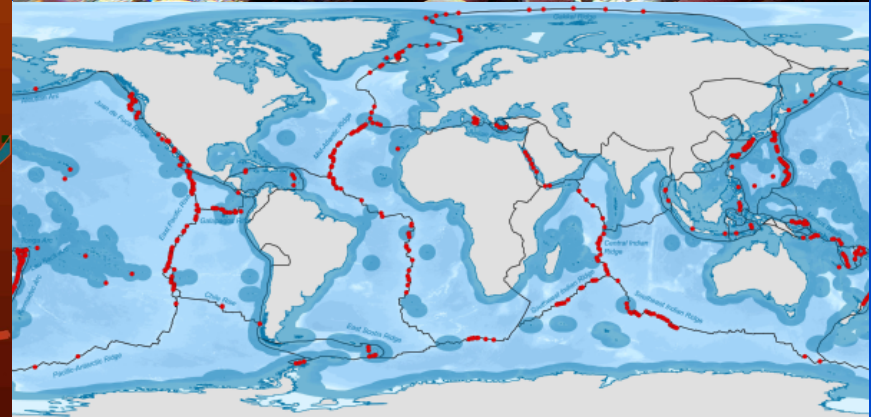
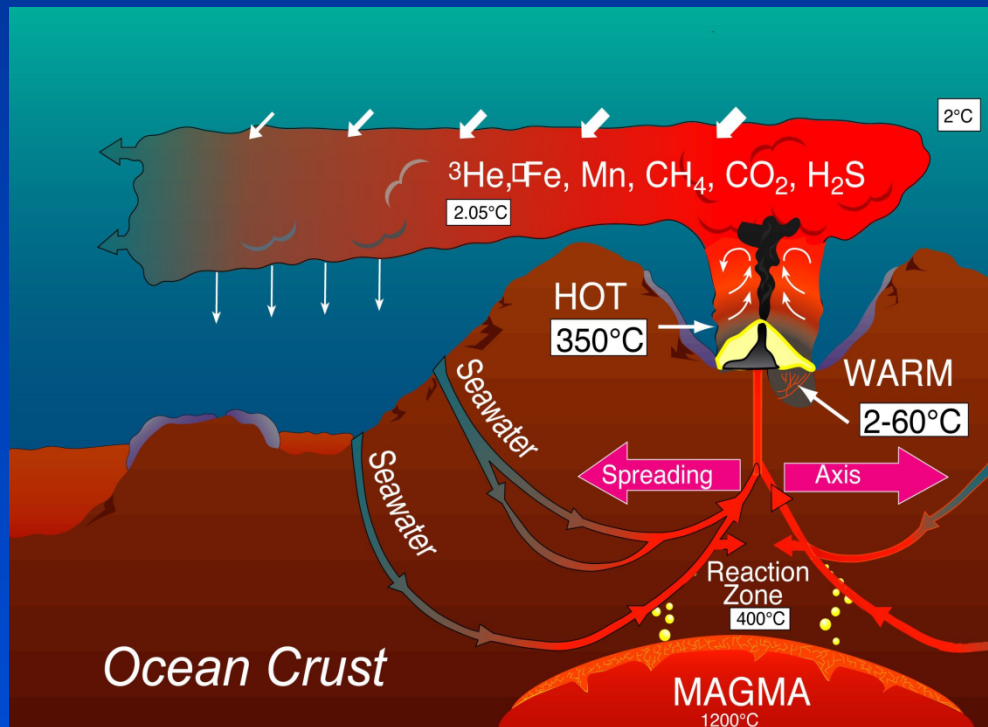


Primary production in oceanic systems

Only a minor fraction of PP in oceans is from chemosynthesis, as in the hydrothermal vents of mid-oceanic ridges, operated by chemoautotroph bacteria

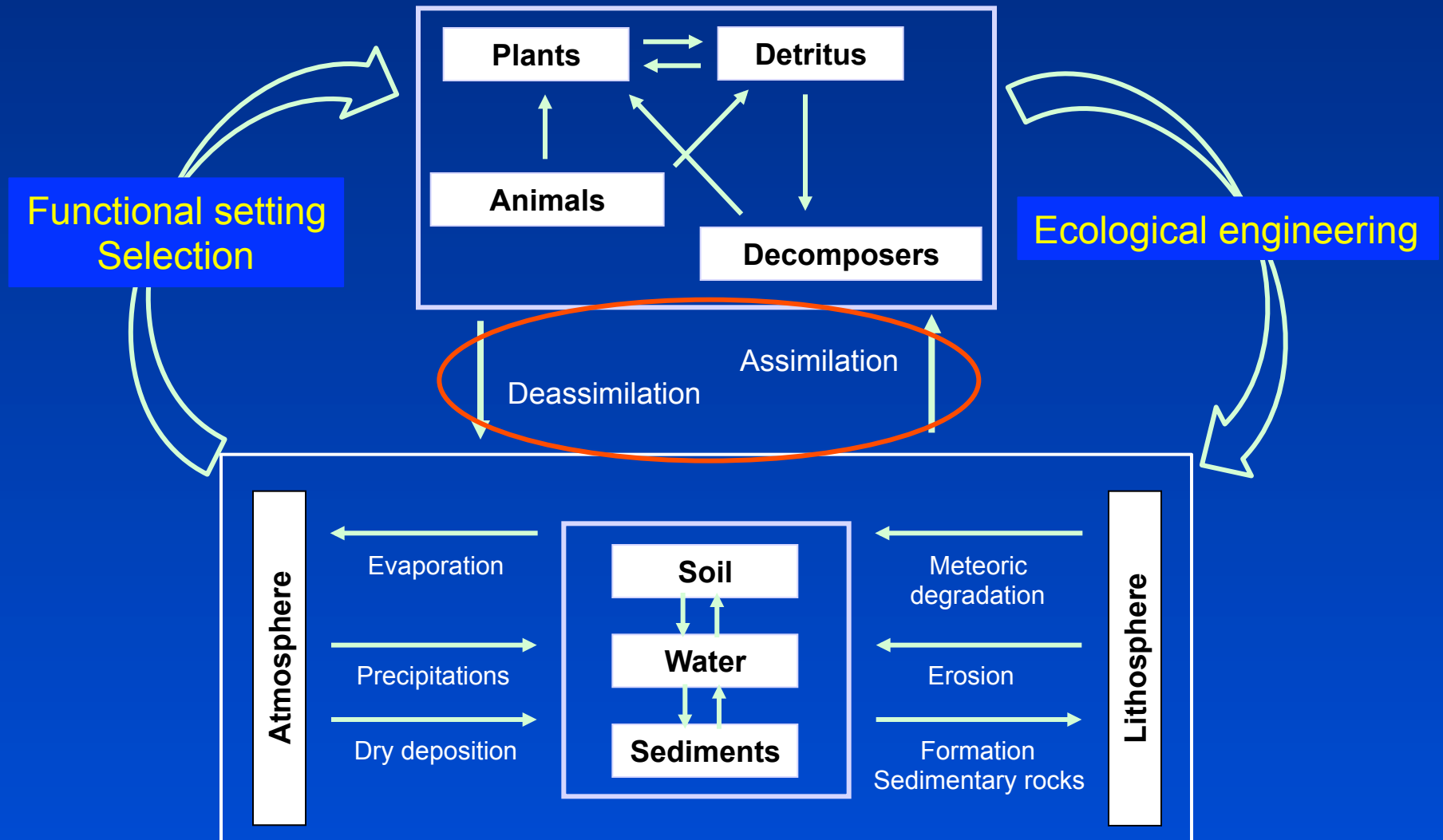


Riftia pachyptila

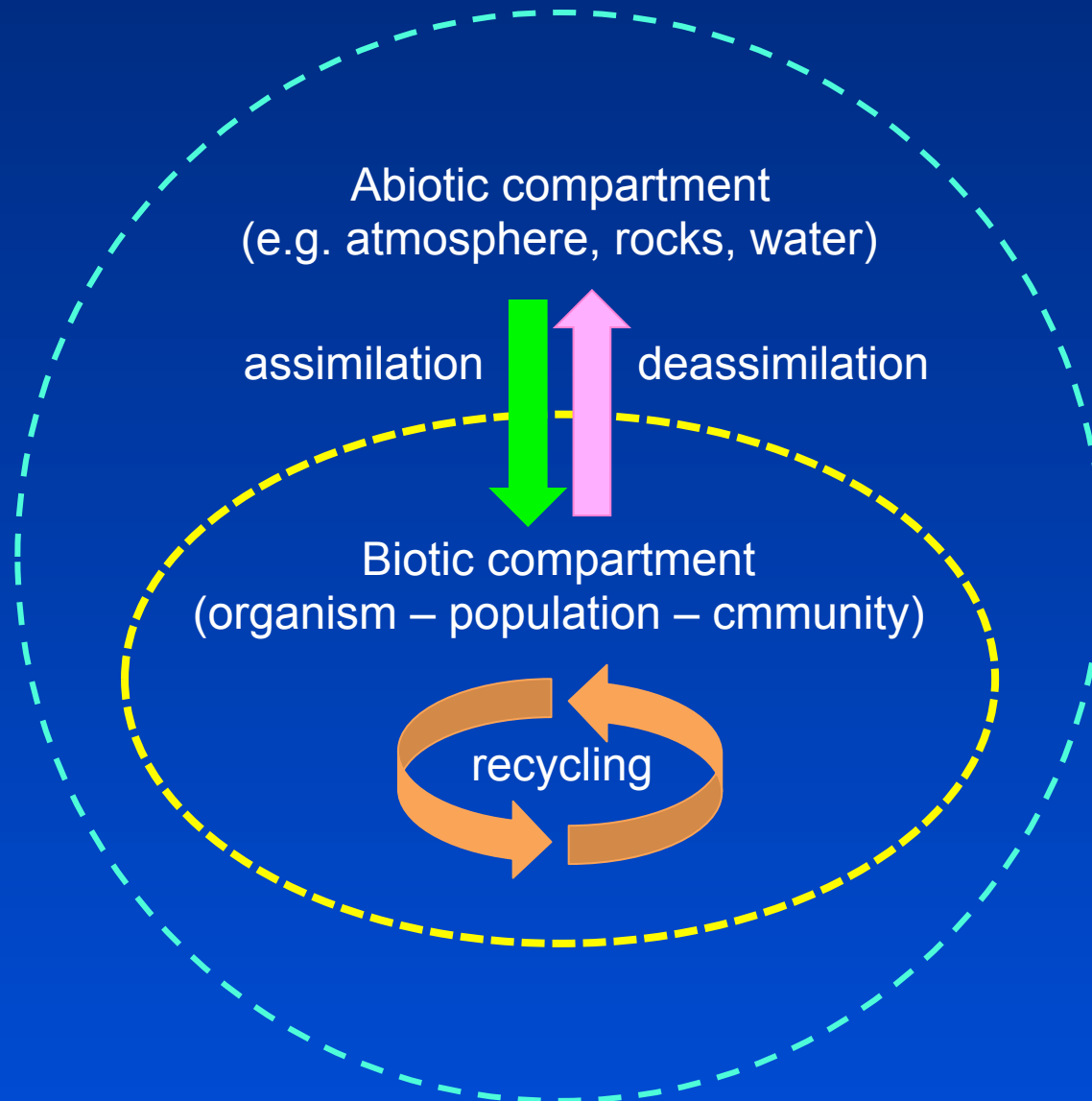


The Ecosystem

The whole set of bio-abiotic compartments and their patterns of interaction



Biogeochemical cycles



Biogeochemical cycles

Which chemical items are subject to biogeochemical cycling

Elements

Macrocomponents (e.g. C, N, O, P, S)

Microcomponents (e.g. Mg, Mn, Fe, Cu)

Molecules

(e.g. H_2O , CH_4 , O_2)

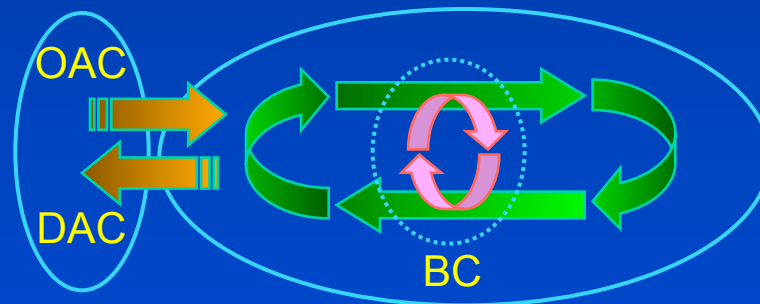
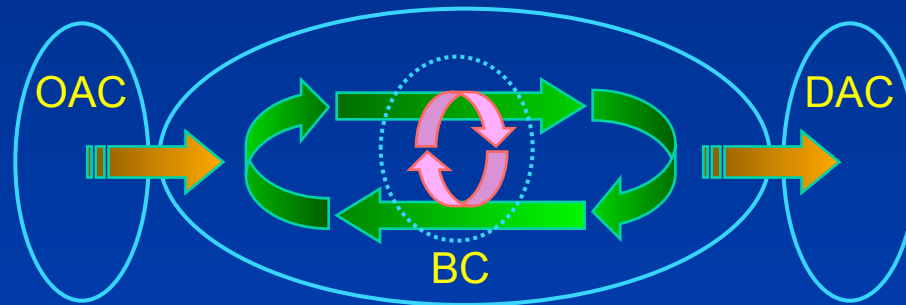
Xenobiotics

(e.g. chlorinated hydrocarbons, organometallic compounds)

Biogeochemical cycles

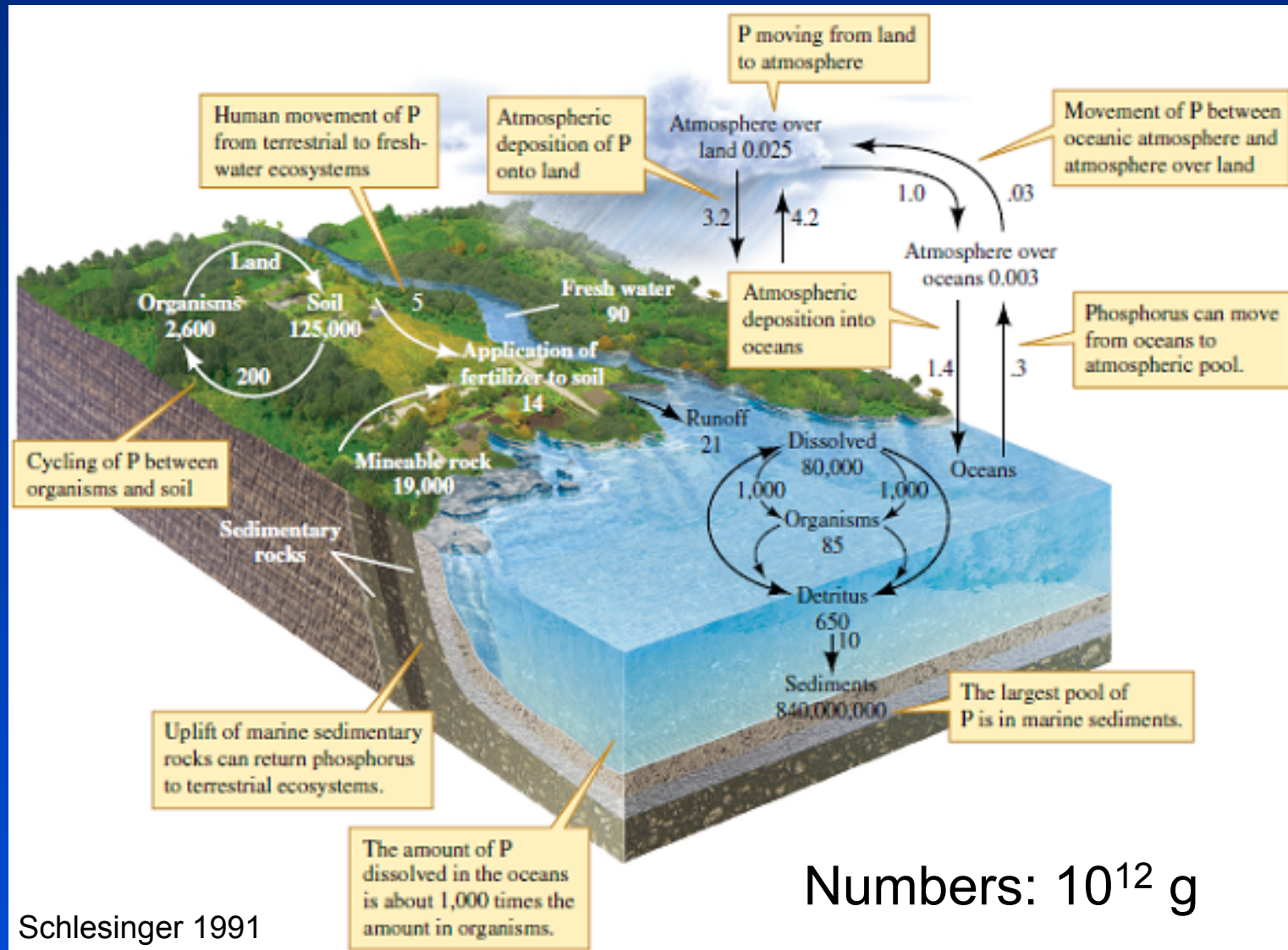
General schemes

Open



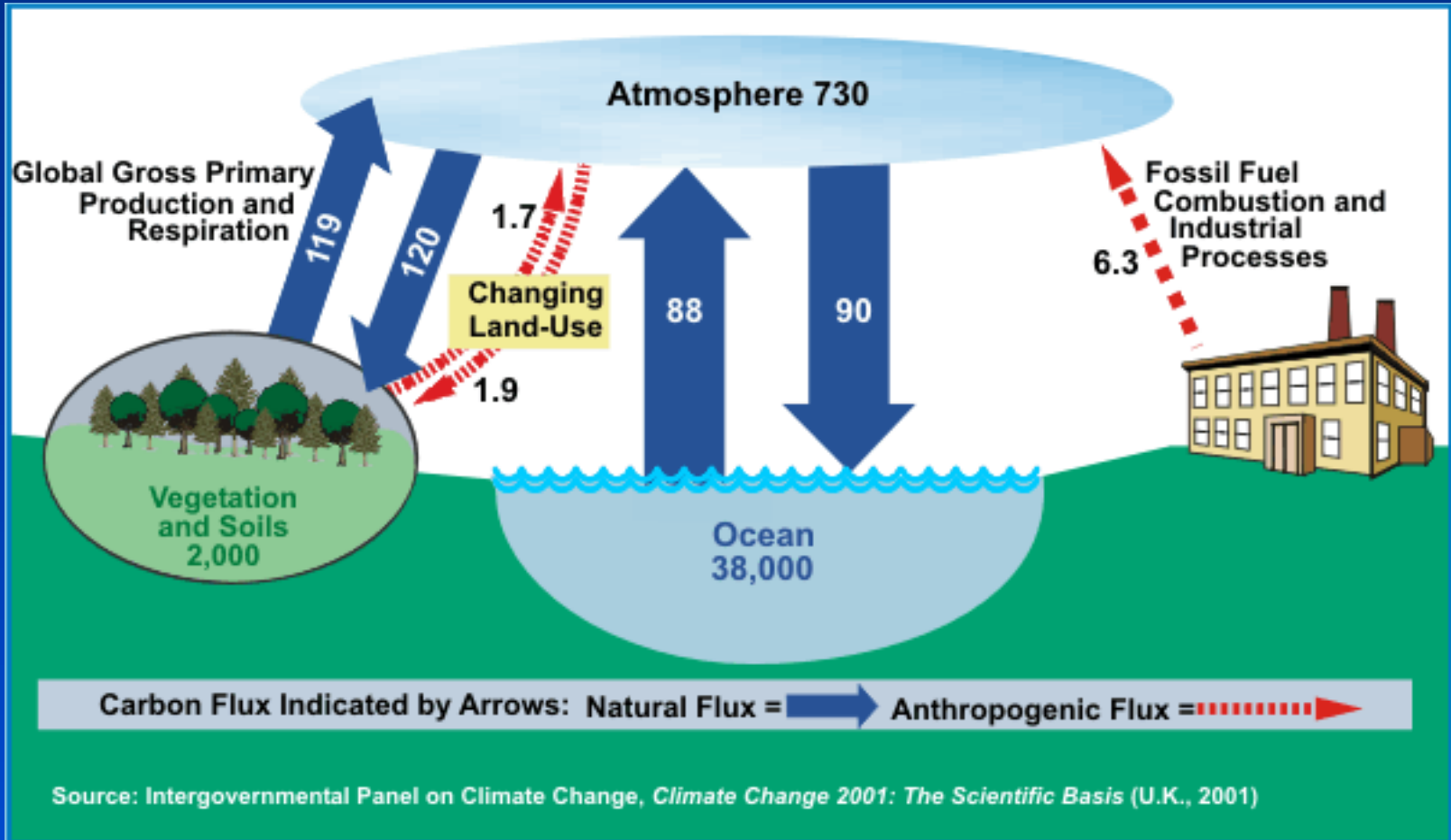
Closed

Phosphorus global cycle (quasi open cycle)



Numbers: 10^{12} g

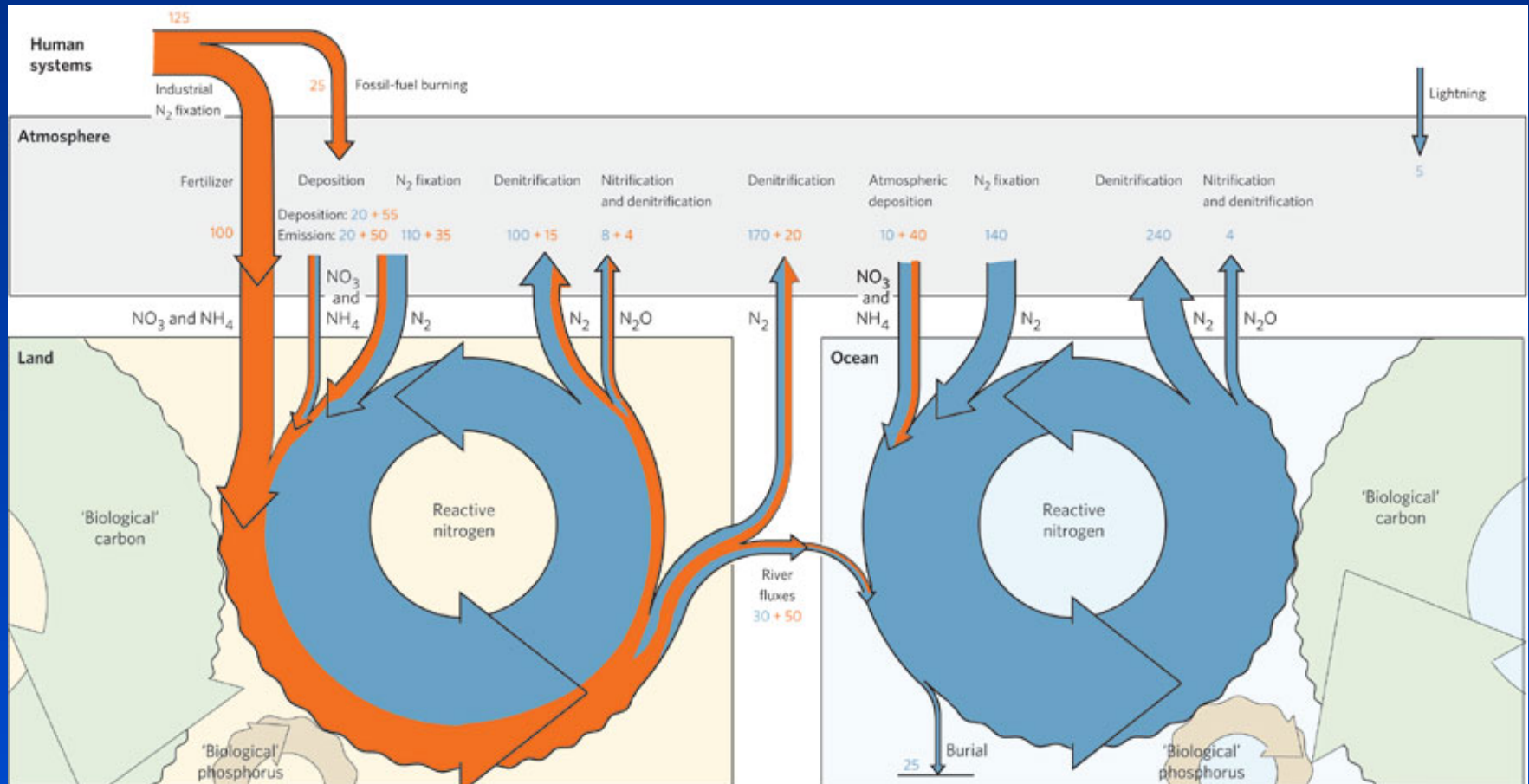
Carbon global cycle (quasi closed cycle) IPCC 2001



Numbers: Gt Yr⁻¹

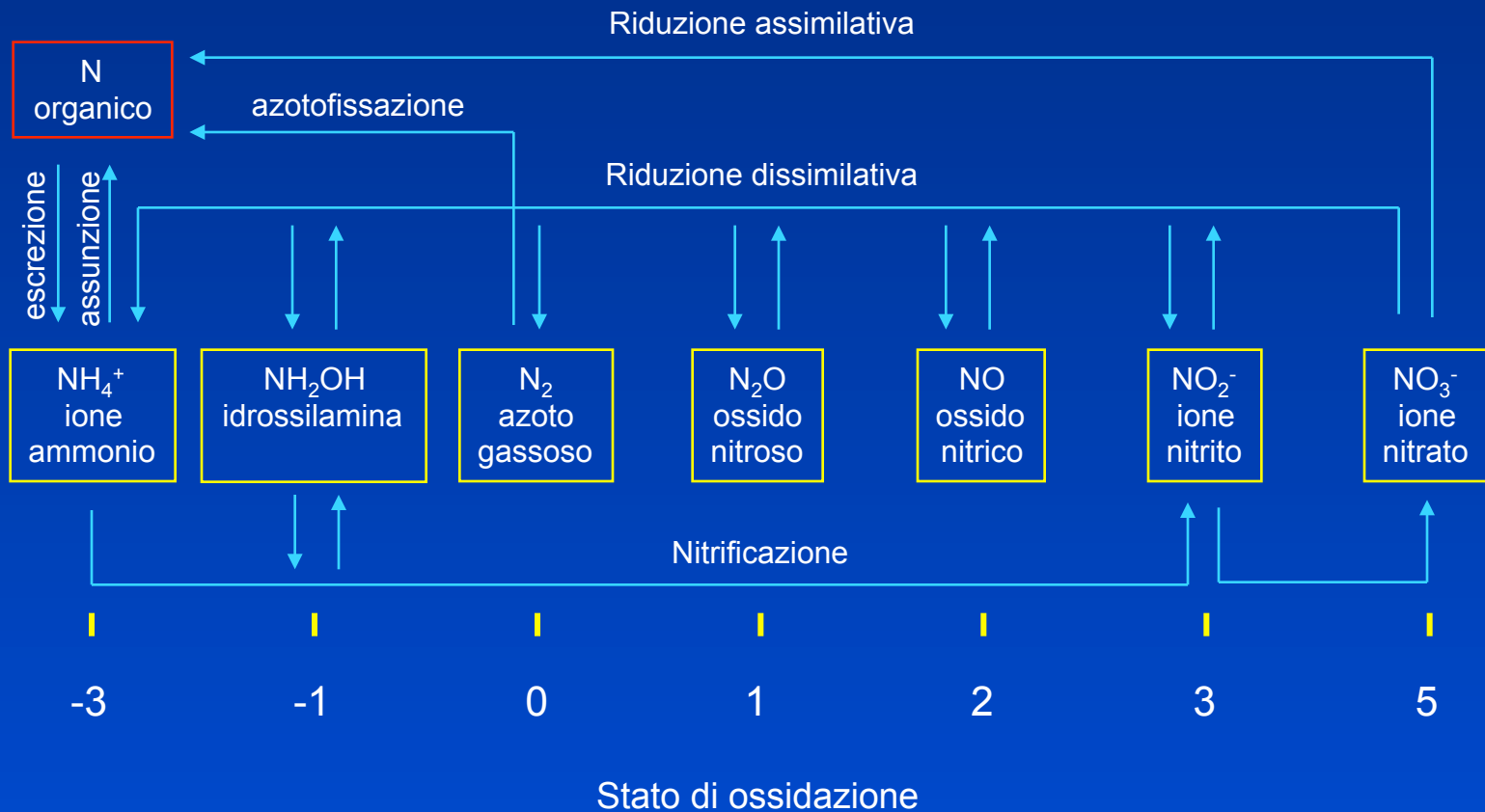
Nitrogen global cycle

Gruber & Galloway 2008 Nature



Fluxes in Tg Yr^{-1}

Redox transformations during Nitrogen global cycle



The role of bacteria in the terrestrial Nitrogen cycle

A. Nitrogen Fixation:

Free-living (e.g. *Azotobacter*) and symbiotic (e.g. *Rhizobium*) bacteria convert atmospheric nitrogen N_2 to ammonia NH_4^+ using the energy from ATP. (In the marine environment N_2 fixation is performed by cyanobacteria).

B. Nitrification:

Several species of soil bacteria convert ammonia NH_4^+ to nitrite NO_2^+ (e.g. *Nitrosomonas*) and nitrite to nitrate NO_3^+ (*Nitrobacter*) when the soil contains high levels of oxygen.

C. Denitrification:

Several species of soil bacteria (e.g. *Pseudomonas*, *Clostridium*) convert nitrate NO_3^+ to free nitrogen N_2 when the soil is waterlogged and thus contains low levels of oxygen. Such bacteria use the nitrate as the final electron acceptor during respiration rather than oxygen. Denitrification depletes the soil of essential nitrate fertilizers.

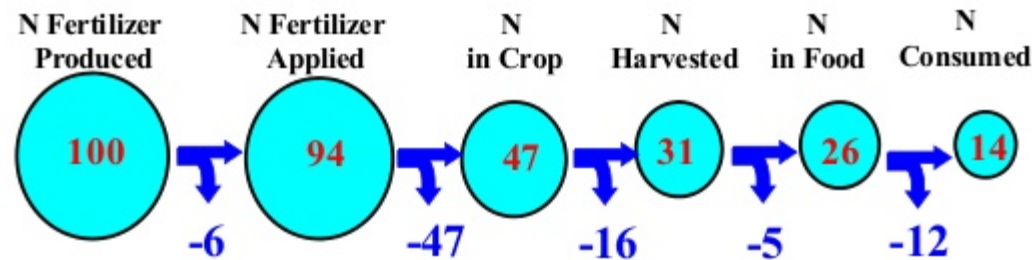
Human effects on Nitrogen cycle

- A. Increase in Nitrogen fixation by use of nitrogen-fixing plants and industrial production of ammonia (Haber-Bosch)
- B. NO_x emission into Atmosphere by civil and industrial fossil fuel burning
- C. Waste water discharge from human and domestic animal populations

Main effects:

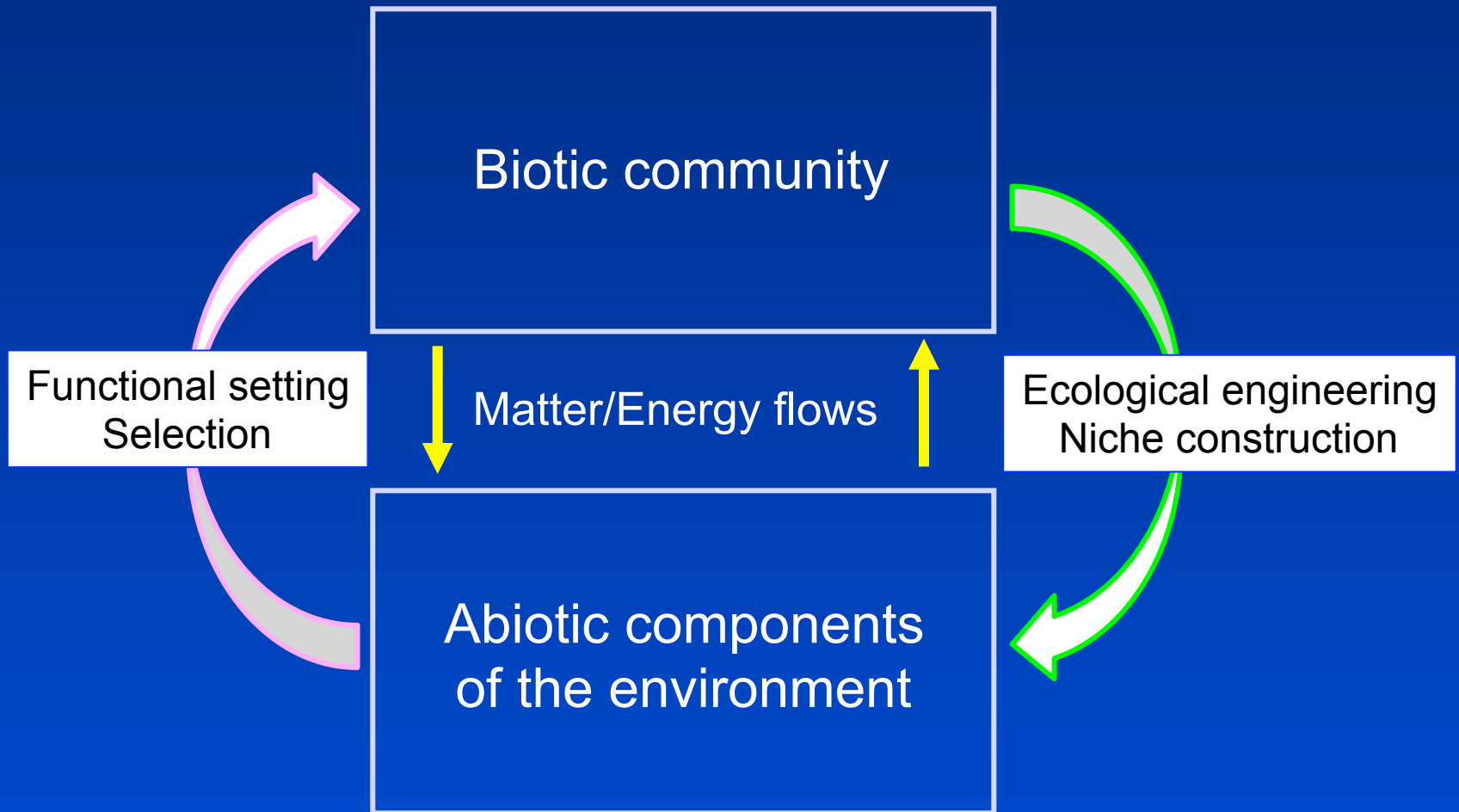
- Eutrophication
- Air, soil and water acidification
- Greenhouse effect (N₂O has a lifetime of 114–120 years, and is 300 times more effective than CO₂ as a greenhouse gas)

The Fate of Haber-Bosch Nitrogen



14% of the N produced in the Haber-Bosch process enters the human mouth.....

Ecological engineering-Niche construction



Examples of niche constructors



Cyanobacteria – Oxyatmosphere



Beaver – Dams/Lakes



Rhinos - Savannah



Scleroactinians – Coral reefs